1 Cu2+|Cu = +0.34 V, H+|H2 = 0 V; Zn2+|Zn = -0.76 V
Placing Cu(s) in 0.2 M HCl(aq), H2 bubbles ...

A should form
B should not form
C further information required

2 Cu2+|Cu = +0.34 V, H+|H2 = 0 V; Zn2+|Zn = -0.76 V
Placing Zn(s) in 0.2 M HCl(aq), H2 bubbles ...

A should form
B should not form
C further information required

3 Cu2+|Cu = +0.34 V, H+|H2 = 0 V; Zn2+|Zn = -0.76 V
Placing Zn(s) in CuSO4(aq), ...

A H2 bubbles should form
B no reaction should occur
C Cu(s) should deposit on the Zn(s)
D further information required
E

4 Cu2+|Cu = +0.34 V, H+|H2 = 0 V; Zn2+|Zn = -0.76 V
Placing Zn(s) coated with Cu(s) in 0.2 M HCl(aq), compared to H2 bubble formation with pure Zn, H2 bubbles should appear ...

A more slowly
B at the same rate
C more quickly