1 E0(Ag+|Ag) = 0.799, E0(Hg2+|Hg) = 0.908, E0(Fe3+|Fe2+) = 0.771, E0(Cd2+|Cd) = -0.40. Which is strongest reducing agent?

A Ag
B Hg
C Fe2+
D Cd

2 E0(Ag+|Ag) = 0.799, E0(Hg2+|Hg) = 0.908, E0(Fe3+|Fe2+) = 0.771, E0(Cd2+|Cd) = -0.40. Which is strongest oxidizing agent?

A Ag+
B Hg2+
C Fe3+
D Cd2+

3 E0(Fe3+|Fe2+) = 0.771, E0(Fe2+|Fe) = -0.409, E0(Cu2+|Cu) = 0.340. What is true about E0(Fe3+|Fe) ?

A E0(Fe3+|Fe) > E0(Cu2+|Cu)
B E0(Fe3+|Fe) > E0(Fe3+|Fe2+)
C E0(Fe3+|Fe) > E0(Fe2+|Fe)
D None of the above