1 The voltage for the cell Zn|Zn2+||Cu2+|Cu is 1.10 V when both ions are at 1 M. As the electrons flow, the voltage will ...

A increase
B stay the same
C decrease

2 The voltage for the cell Zn|Zn2+||Cu2+|Cu is 1.10 V when both ions are at 1 M. If the Cu2+ concentration is increased to 10 M, the voltage will ...

A increase
B stay the same
C decrease

3 The voltage for the cell Zn|Zn2+||Cu2+|Cu is 1.10 V when both ions are at 1 M. If the Zn2+ concentration is increased to 10 M, the voltage will ...

A increase
B stay the same
C decrease

4 The voltage for the cell Zn|Zn2+||Cu2+|Cu is 1.10 V when both ions are at 1 M. If the Cu2+ and Zn2+ concentrations are each increased to 10 M, the voltage will ...

A increase
B stay the same
C decrease

5 The voltage for the cell Zn|Zn2+||Cu2+|Cu is 1.10 V when both ions are at 1 M. If the Cu2+ and Zn2+ concentrations are each decreased to 0.01 M, the voltage will ...

A increase
B stay the same
C decrease