

1 What is the shape of  $\text{H}_2\text{S}$ ?

- A Linear
- B Bent
- C Trigonal pyramid
- D Tetrahedral
- E None of the above

2 What is the shape of  $\text{NH}_4^+$ ?

- A Linear
- B Bent
- C Trigonal pyramid
- D Tetrahedral
- E None of the above

3 What is the shape of  $\text{NH}_3$ ?

- A Linear
- B Bent
- C Trigonal pyramid
- D Tetrahedral
- E None of the above

4 What is the shape of  $\text{NH}_2^-$ ?

- A Linear
- B Bent
- C Trigonal pyramid
- D Tetrahedral
- E None of the above

5 What is the shape of  $\text{CH}_4$ ?

- A Linear
- B Bent
- C Trigonal pyramid
- D Tetrahedral
- E None of the above

6 What is the shape of  $\text{BF}_3$ ?

- A Linear
- B Bent
- C Trigonal pyramid
- D Tetrahedral
- E None of the above

- 7 The shape of  $\text{NH}_3$  is trigonal pyramid. One proton is removed from the N of  $\text{NH}_3$ . What is the shape of the resulting molecular ion?
- A Trigonal pyramid
  - B Trigonal planar
  - C Bent
  - D None of the above
- 8 The shape of  $\text{NH}_3$  is trigonal pyramid. One proton is removed from one of the H's of  $\text{NH}_3$ . What is the shape of the resulting molecular ion?
- A Trigonal pyramid
  - B Trigonal planar
  - C Bent
  - D None of the above