

Lecture 6 (9/17/25)

- Reading: Ch3; 83–87, 89–90
Ch1; Fig 7
Ch9; 313–314

- Homework #6

NEXT

- Review session

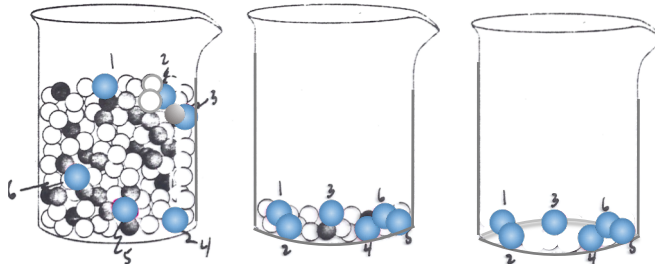
OUTLINE

Protein Purification

- A. Introduction; what is the basis
- B. Goals; Specific Activity
- C. Methods
 - 1. Centrifugation
 - a. Differential
 - b. Isopycnic
 - 2. Precipitation
 - a. Salting-out; ammonium sulfate
 - b. dialysis
 - 3. Chromatography
 - a. Gel filtration
 - b. Ion exchange
 - c. Affinity
- D. Summary

Protein Purification

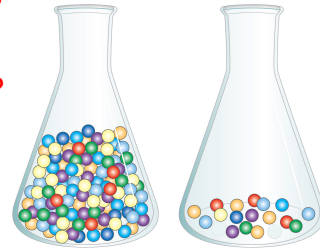
$$\bullet \text{ Specific Activity} = \frac{\text{Activity of YFP}}{\text{Total protein}}$$



How do you measure Specific Activity?

Figure 6-3 Activity versus specific activity. The difference between these two terms can be illustrated by considering two jars of marbles. The jars contain the same number of blue marbles (representing an unknown protein), but different amounts of marbles of other colors. If the marbles are taken to represent proteins, both jars contain the same activity of the protein represented by the blue marbles. The second jar, however, has the higher specific activity because here the blue marbles represent a much higher fraction of the total.

What is the Yield?
(of red proteins?)



From [protein] (mg/mL) How do you measure specific activity? From [activity] (U/mL)

Purification of a hypothetical protein

Procedure or step	Fraction volume (mL)	Total protein (mg)	Activity (units)	Specific activity (units/mg)	Yield (%)
1. Crude cellular extract	1,400	10,000	100,000	10	100
2. Precipitation	280	3,000	96,000	32	96
3. Ion-exchange chromatography	90	400	80,000	200	80
4. Size-exclusion chromatography	80	100	60,000	600	60
5. Affinity chromatography	6	3	45,000	15,000	45

These are the two important criteria of purifications

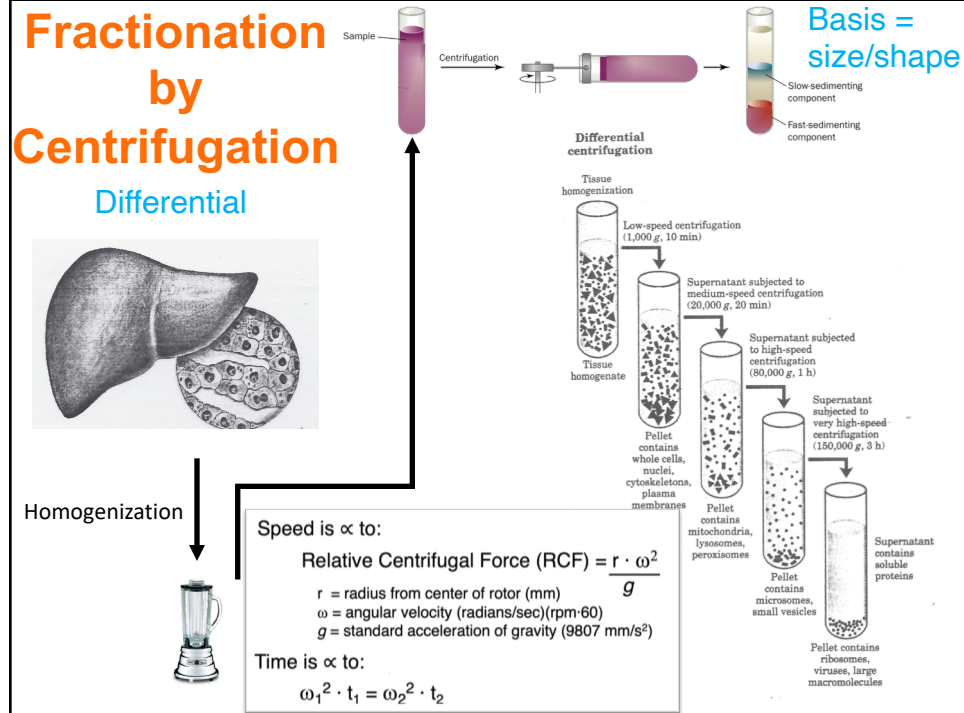
-If protein was "pure" after step #5, what would the Specific Activity be after you performed a step #6?

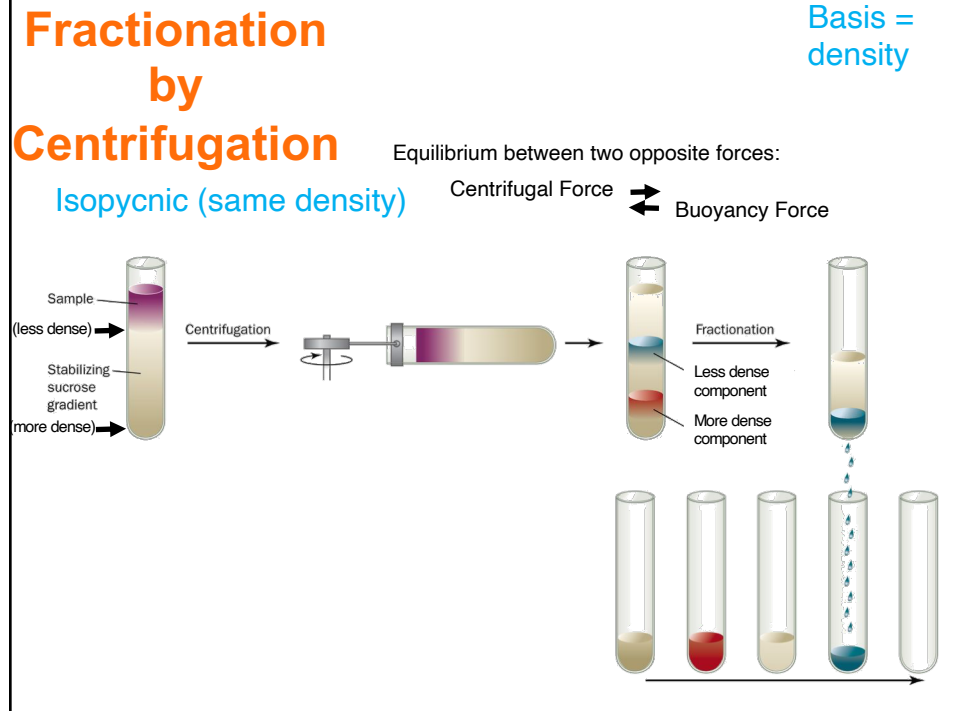
-What is the Yield?

* All data represent the status of the sample after the procedure indicated in the first column has been carried out.

Protein Purification Procedures

Basis	Procedure	Covered
Hydrodynamics (size, shape, density)	Gel filtration <u>Chromatography</u>	Lab
	SDS-PAGE	Lab
	Centrifugation	Lab
Charge	Ion exchange <u>Chromatography</u>	
	Isoelectric focusing	
	Native electrophoresis	Lab
Solubility	Salting out	
	Organic extraction	
	Hydrophobic interaction	
	Chromatography	
Binding Specificity	Affinity <u>Chromatography</u>	Lab



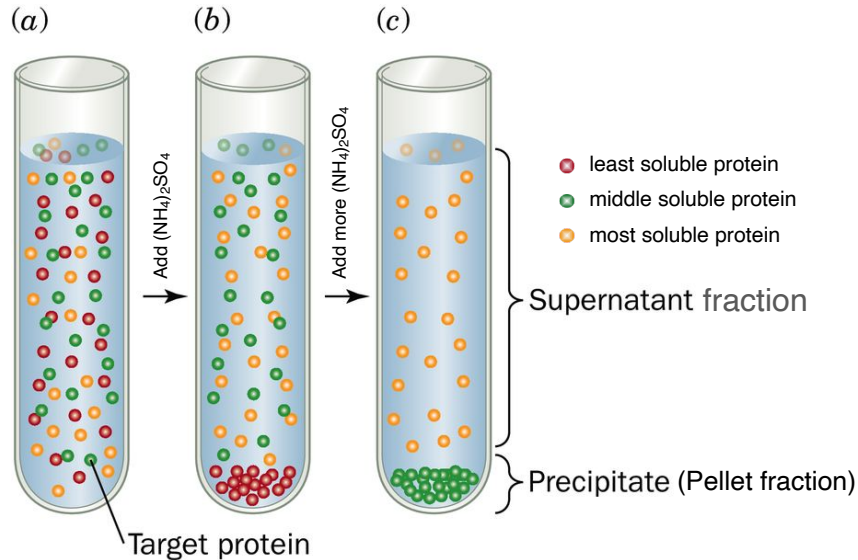


Protein Purification Procedures

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Ammonium Sulfate Fractionation: Fractionation by Salting Out

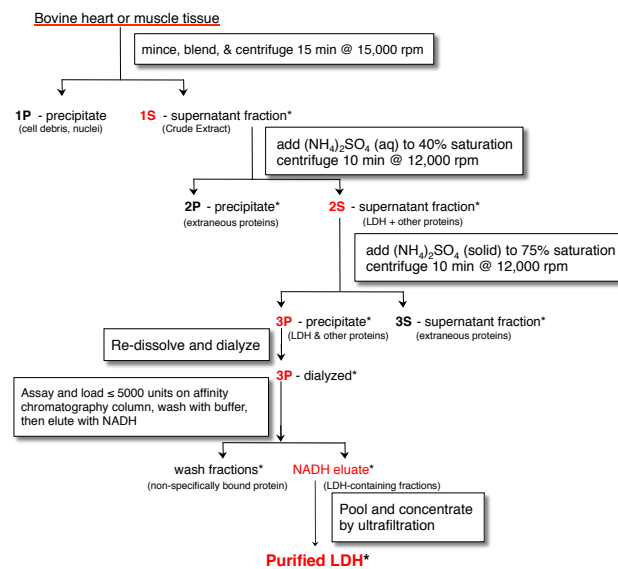
Basis = solubility



Purification in Laboratory begins with Fractionation by Salting Out

Basis = solubility

FLOW CHART FOR LDH PURIFICATION



Proteins are Least Soluble at Their Isoelectric Point

Basis =
solubility

TABLE 5-3 Isoelectric Points of Several Common Proteins

Protein	pI
Pepsin	<1.0
Ovalbumin (hen)	4.6
Serum albumin (human)	4.9
Tropomyosin	5.1
Insulin (bovine)	5.4
Fibrinogen (human)	5.8
γ -Globulin (human)	6.6
Collagen	6.6
Myoglobin (horse)	7.0
Hemoglobin (human)	7.1
Ribonuclease A (bovine)	9.4
Cytochrome <i>c</i> (horse)	10.6
Histone (bovine)	10.8
Lysozyme (hen)	11.0
Salmine (salmon)	12.1

Cation $\rightarrow$ Zwitterion $\rightarrow$ Anion
pH at: <pI pI >pI

Notice that
between 7.1 –
9.4 most of
these proteins
are charged

Acidic proteins

Example of this
at the end

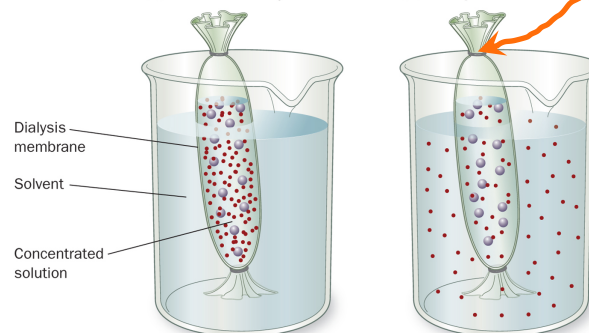
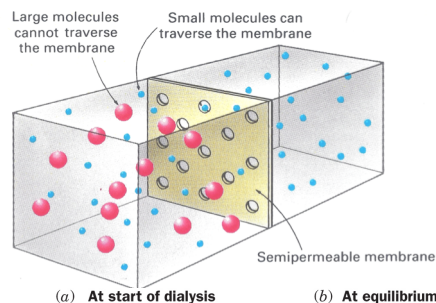
Basic proteins

If you are performing a salting-out procedure prior to most other procedures, you will need to remove the Molar concentrations of salt. How?

Dialysis: Diffusion of Solutes

Basis =
solubility

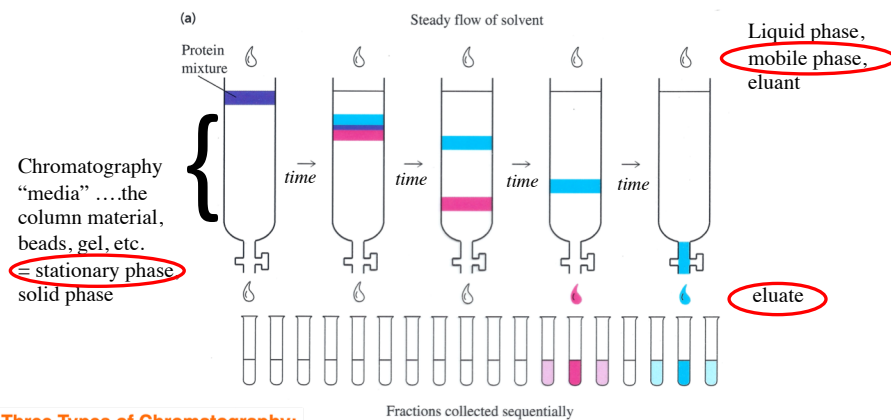
MWCO=molecular
weight cutoff



Protein Purification Procedures

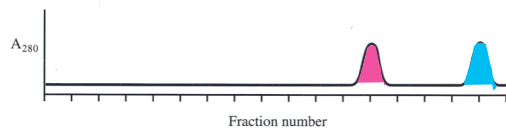
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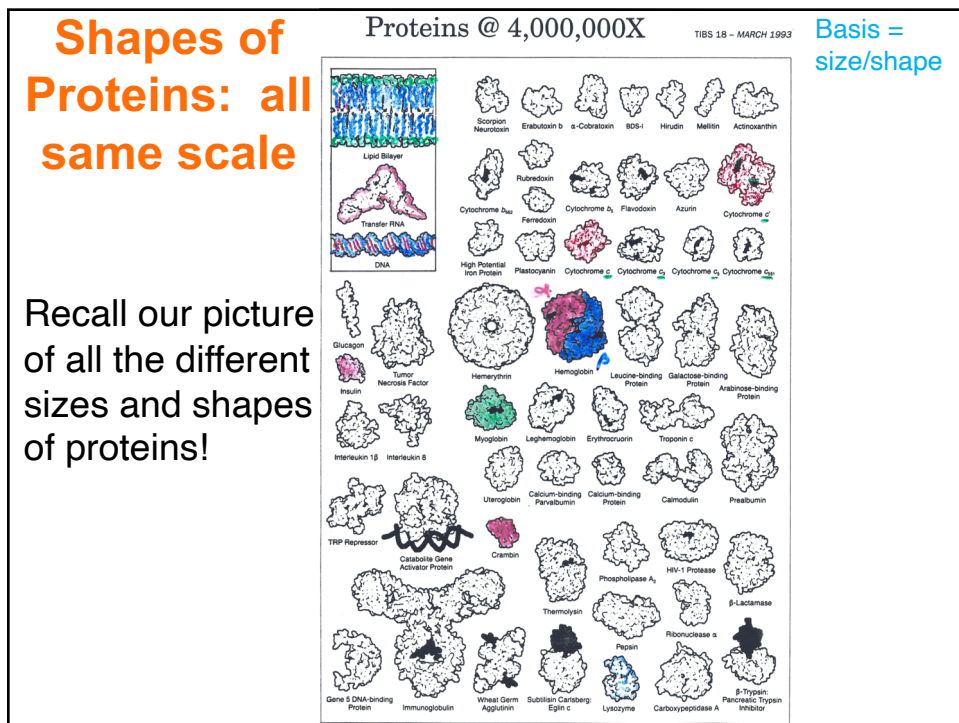
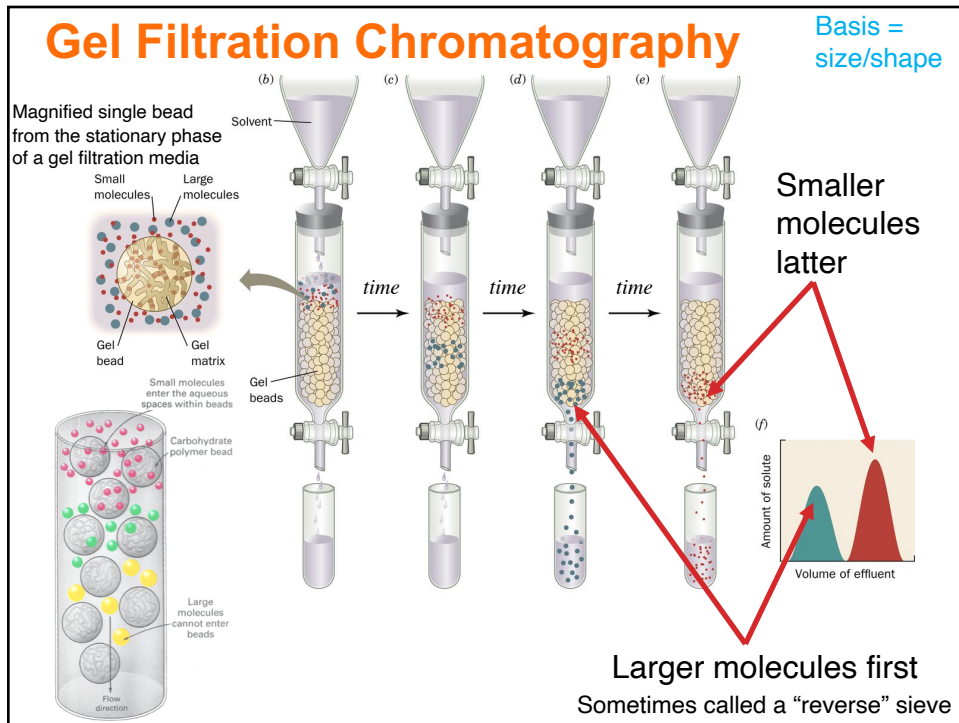
Basic Chromatography



Three Types of Chromatography:

- 1) Gel Filtration (size/shape)
- 2) Ion exchange (charge)
- 3) Affinity (function)





Protein Purification Procedures

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	Hydrophobic interaction Chromatography	
Binding Specificity	Affinity <u>Chromatography</u>	Lab

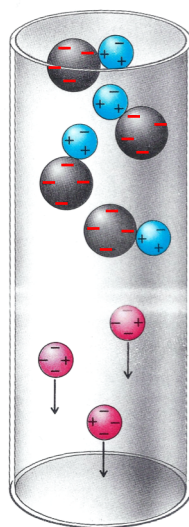
Ion Exchange Chromatography

Basis =
charge

At certain pH values, molecules will have a net charge:

Cation → Zwitterion → Anion
pH at: < pI pI > pI

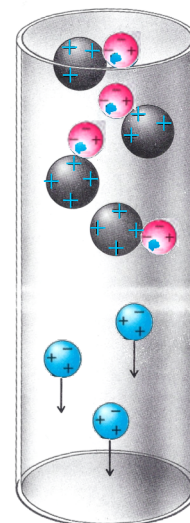
- Positive (+) when $\text{pH} < \text{pI}$
- Negative (-) when $\text{pH} > \text{pI}$



Positively charged protein binds to **negatively** charged bead

Negatively charged protein flows through

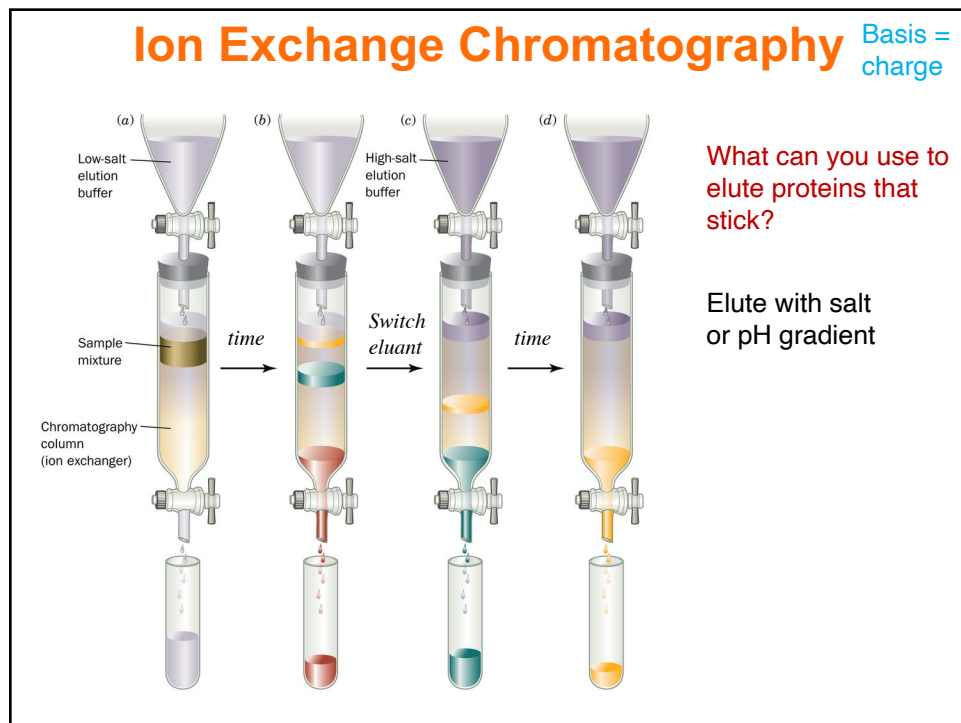
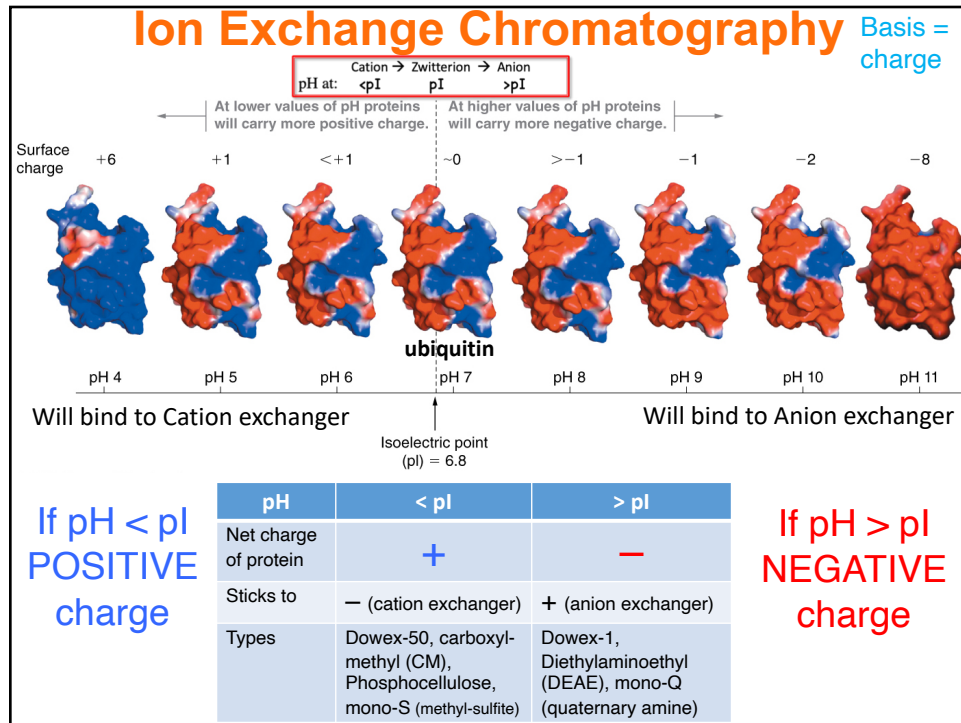
Negatively charged particles/beads = **cat-ion exchange**



Negatively charged protein binds to **positively** charged bead

Positively charged protein flows through

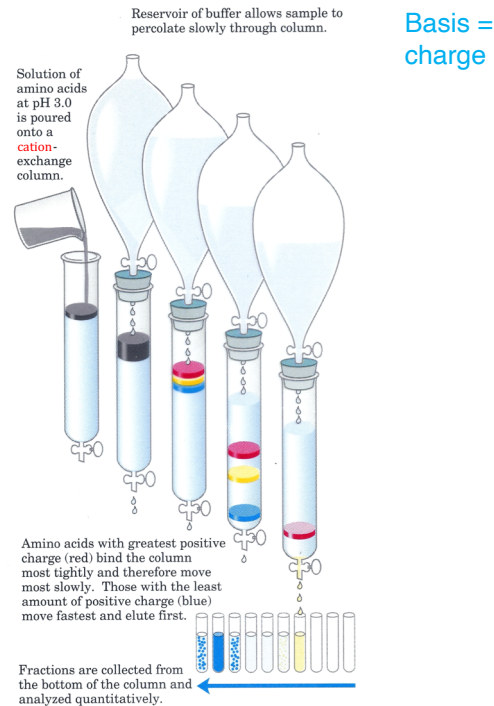
Positively charged particles/beads = **anion exchange**



Ion Exchange Chromatography: EXAMPLE of an amino-acid mixture

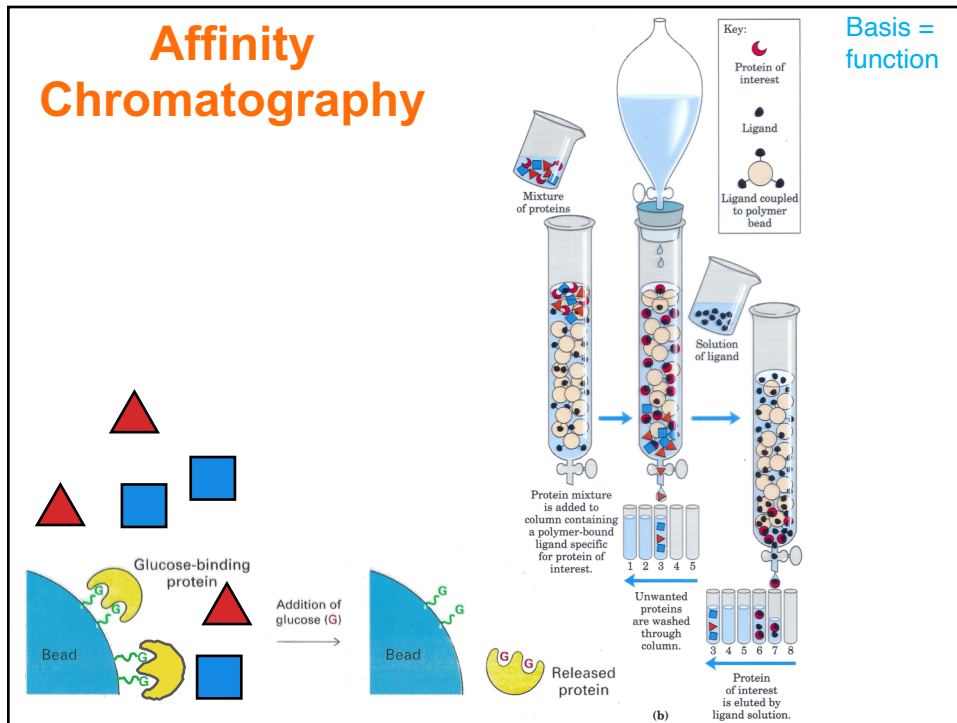
If this were a mixture of G, K, and D, what would the order of elution be?

1st = D, 2nd = G, 3rd = K



Protein Purification Procedures

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Affinity Chromatography

Basis = function

Biotechnology (recombinant DNA technology) has revolutionized protein purification.

At the level of the DNA sequence, the DNA sequence encoding such binding proteins or "tags" can be "fused" to the sequence encoding YFP. In this way, a chimeric protein is produced that has the binding function, which allows the use of affinity chromatography.

<u>Common "tags" are:</u>	<u>Column beads have attached:</u>
Maltose-binding protein	Maltose
Chitin-binding protein	Chitin
Glutathione-S-transferase	Glutathione (γ -Glu-Cys-Gly)
His-His-His-His-His-His	Ni-chelate

Protein Purification Procedures

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Purification of Myoglobin (Mb)

Step	Total Protein (mg)	Mb (μmol)	Specific Activity (μmol Mb/mg Total Protein)	% Yield	Overall Fold Purification
1. Crude extract	1550	0.75		100	1
2. DEAE-cellulose chromatography	550	0.35			1.3
3. Affinity chromatography	5.0	0.28			117

Calculations:

Specific Activity	Yield	Fold Purification
$0.75 \div 1550 = 0.00048$		
$0.35 \div 550 = 0.00064$	$0.35 \div 0.75 = 0.47$	$0.00064 \div 0.00048 = 1.3$
$0.28 \div 5.0 = 0.056$	$0.28 \div 0.75 = 0.37$	$0.056 \div 0.00048 = 117$

Purification of a hypothetical protein

Procedure or step	Fraction volume (ml)	Total protein (mg)	Activity (units)	Specific activity (units/mg)	Fold increase in SA	Yield (%)	% loss in Yield
1. Crude cellular extract	1,400	10,000	100,000	10	3x	100	4%
2. Precipitation	280	3,000	96,000	32		96	
3. Ion-exchange chromatography	90	400	80,000	200	6x	80	17%
4. Size-exclusion chromatography	80	100	60,000	600	3x	60	25%
5. Affinity chromatography	6	3	45,000	15,000	25x	45	25%

* All data represent the status of the sample *after* the procedure indicated in the first column has been carried out.

Calculate fold increase in SA for each step → helps determine if step is effective.

Which is the best step?step 5

Which is the worst step?step 2 or step 4

Look at yield..... Calculate fraction (%) of YFP lost at each step.

Which of step 2 or step 4 resulted in loss of more YFP?step 4