# Lecture 21 (11/5/25)

TODAY
•Reading: Ch7; 229-235

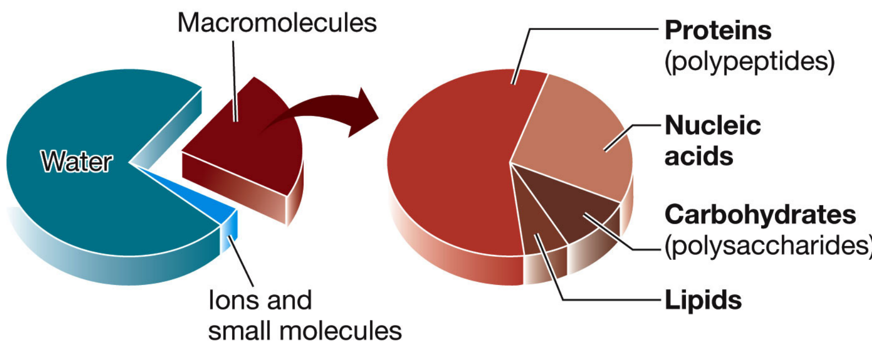
**NEXT**
•Reading: Ch7; 236-241

## Carbohydrates

A. Definition
B. Roles
C. Monosaccharides-Chemistry
   1. Chirality
      a. One or more asymmetric carbons
      b. Linear and ring forms
   2. Derivatives: the chemistry of carbohydrates
      a. Oxidation
         i. C1
         ii. C6
      b. Reduction
         i. C1/C2
         ii. Other carbons
      c. Ester formation
      d. Amino sugars
   3. Polymerization
      a. The Glycosidic Bond
      b. Non-covalent bonds in macro-molecular structure
D. Oligosaccharides
   1. Glycoproteins & glycolipids
   2. O-linked
   3. N-linked
   4. Sequence determination-ABO
E. Polysaccharides
   1. Polymers of glucose
   2. Polymers of disaccharides

# Carbohydrates

## Carbohydrates




## Carbohydrates

### Definition

- Carbo-Hydrate: have formula $C_n(H_2O)_n$ (for $n \geq 3$)
- The precursor-macromolecule relationship is:
  - Monosaccharide–polysaccharide (or oligosaccharide)
- Carbohydrates are everywhere (ubiquitous) and versatile in function; fulfill a variety of functions.
  - Can be covalently linked with proteins and lipids; are intimately involved in nucleic acids

| | ROLES | Monosaccharide | Polysaccharide |
|---|---|---|---|
| 1. | Energy source/storage | glucose, fructose, etc. | Starch, glycogen |
| 2. | Structure | glucose, *glycerol* | Cellulose, chitin, lipids & membranes |
| 3. | Information | ribose (nucleotides) | Nucleic acids |
| 4. | Recognition | many | Glycolipids & glycoproteins |

# Carbohydrates

## The 4 S's

**Size** — Range from as small as glyceraldehyde ($M_w$ = 90 g/mol) to as large as amylopectin ($M_w$ > 200,000,000 g/mol)

**Shape** — Depends on size and glycosidic bond

**Solubility** — Very polar, very soluble, until large polymers

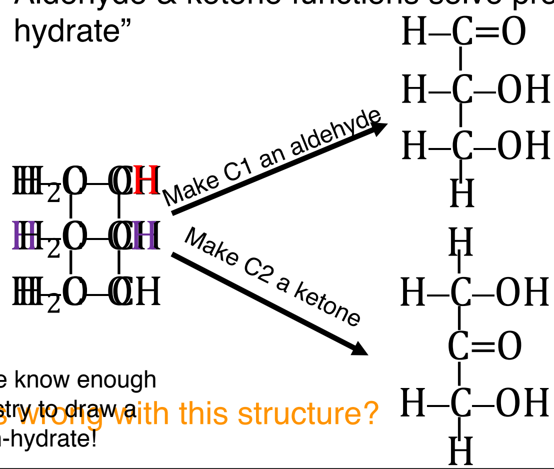
**Stability** — Stable due to glycosidic bond

# Carbohydrates

### Monosaccharides

$C_3(H_2O)_3$

- Basic nomenclature:
  - Use the suffix "-ose" — triose, tetrose, pentose, hexose
  - Aldehyde & ketone functions solve problem of a "carbo-hydrate"




= aldoses

aldotriose (triose), aldotetrose(tetrose), aldopentose, aldohexose

= ketose

Ketotriose (triulose), ketotetrose (tetrulose)… (pentulose, hexulose).

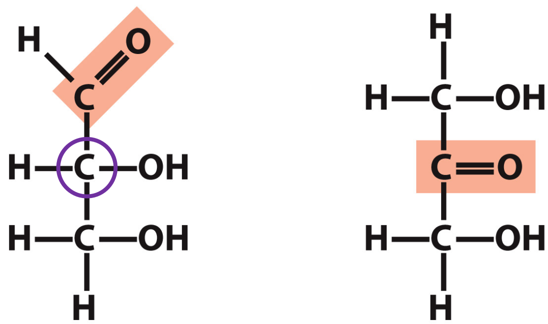
OK, we know enough chemistry to draw a carbon-hydrate!

What is wrong with this structure?

# Carbohydrates

## Monosaccharides

- An aldose is a carbohydrate with aldehyde functionality.
- A ketose is a carbohydrate with ketone functionality.



Glyceraldehyde, an aldotriose

Dihydroxyacetone, a ketotriose

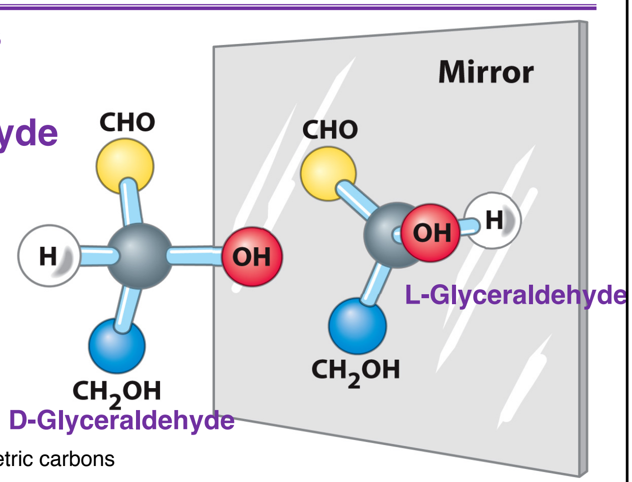
Are there any chiral carbons?

Figure 7-1a
*Lehninger Principles of Biochemistry*, Seventh Edition


# Carbohydrates

## Monosaccharides

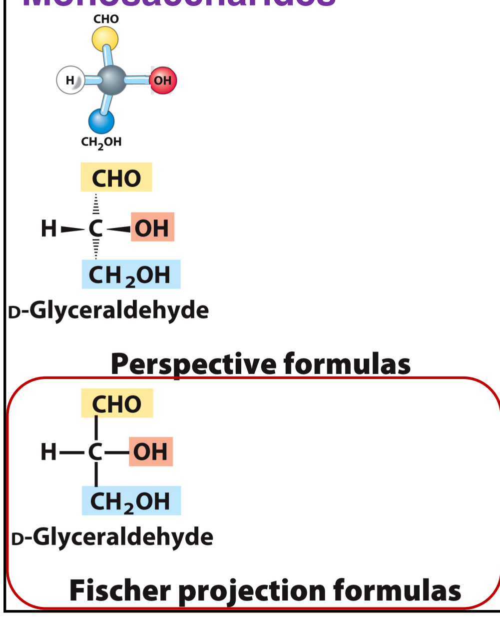
Glyceraldehyde




- Chemical Features:
  - Chirality
    - One or more asymmetric carbons
    - Linear and ring forms
  - Derivatives: the chemistry of carbohydrates
  - Polymerization
    - The Glycosidic Bond
    - Non-covalent bonds in macro-molecular structure

# Carbohydrates

## Monosaccharides




### Fischer projections

- Vertical bonds are between carbons, with highest oxidation state at the top, AND project away from you.
- Horizontal bonds are pointing toward you.
- If hydroxyl is on the left; its **L**
- If hydroxyl is on the right; its **D**
- Here D=R & L=S

It turns out that the L form of glyceraldehyde is called L because it is "levorotary," meaning it will rotate plane-polarized light to the left, or counter-clockwise. D rotates "dextrorotary."

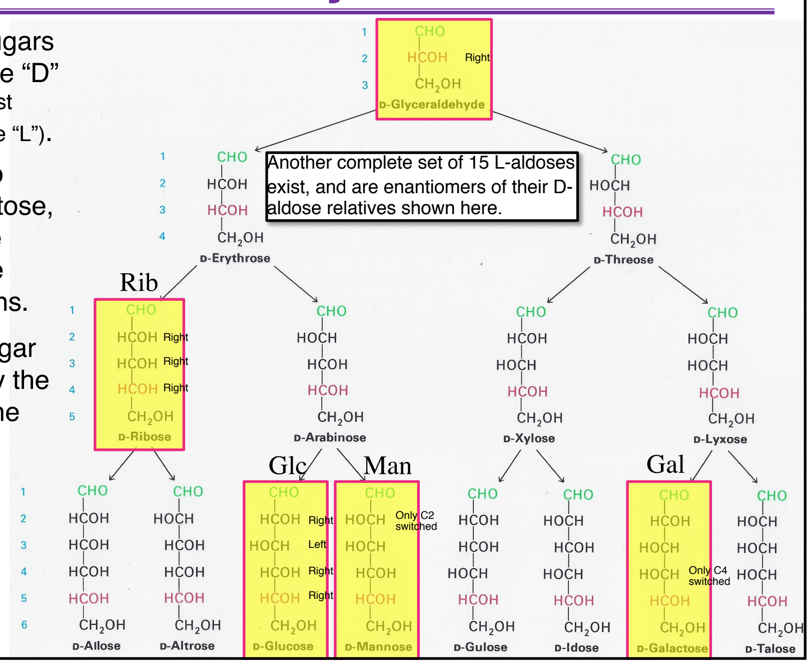
# Carbohydrates

GREEN =functional group
RED=highest numbered chiral carbon

Nearly all sugars in biology are "D" (sort of like most amino acids are "L").

As you go to tetrose, pentose, etc., you are adding more chiral carbons.

A D- or L-sugar is defined by the chirality of the highest numbered carbon.



## Carbohydrates

### D-ketoses

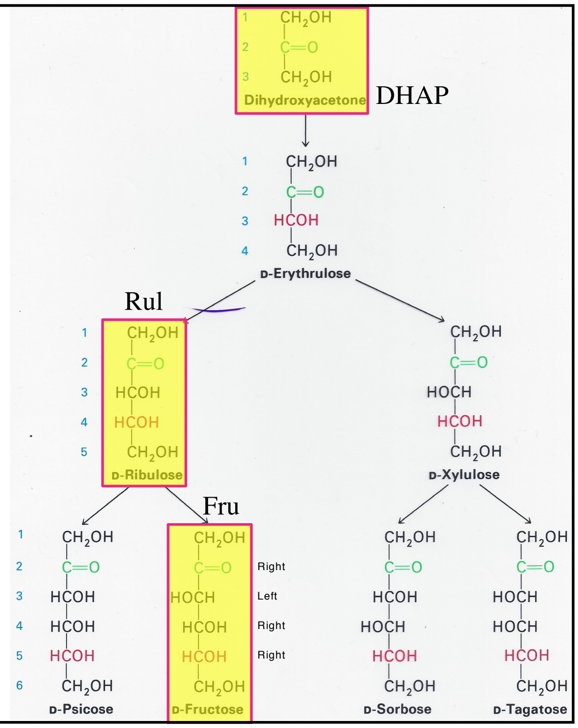
The first chiral sugar is a tetrulose.

There is also another complete set of 7 L-ketoses exist and are enantiomers of their D-ketose relatives shown here.

DHAP is the ketose form of glyceraldehyde

Ribulose is the ketose form of ribose

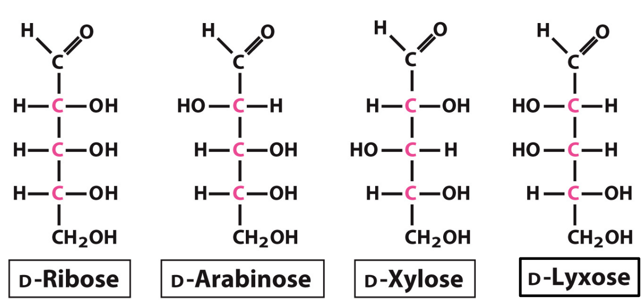
Fructose is the ketose form of glucose




# Carbohydrates

## Monosaccharides: Stereoisomer Nomenclature

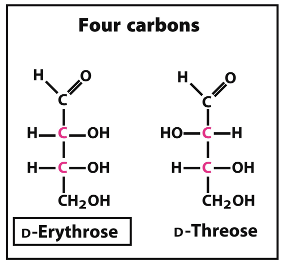
- Enantiomers
  - stereoisomers that are nonsuperimposable **complete** mirror images
  - Example: D-sugars & L-sugars
- Diastereomers
  - stereoisomers that are not complete mirror images
  - Diastereomers have different physical properties (e.g., water solubility)
  - Example:



# Carbohydrates

## Monosaccharides: Stereoisomer Nomenclature

- Epimers
  - Epimers are stereoisomers that differ at only **one** chiral center
  - Epimers are diastereomers; diastereomers have different physical properties (i.e., water solubility, melting temp)
  - Example:

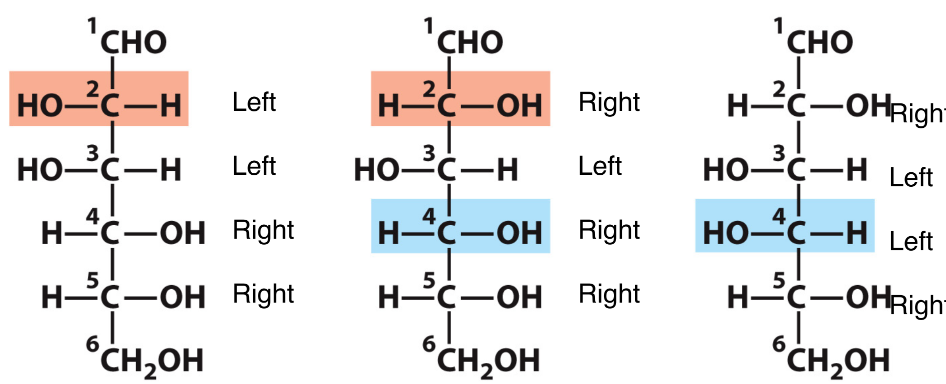



  - Glc & Man (2 epimer); Glc & Gal (4 epimer); Rib & Ara (2 epimer); Rib & Xyl (3 epimer)
- Anomers
  - Anomers have different chirality at carbon involved in ring formation

# Carbohydrates

## Monosaccharides: Stereoisomer Nomenclature

- D-Mannose and D-galactose are both epimers of D-glucose.
- D-Mannose and D-galactose vary at more than one chiral center and are diastereomers, but <u>not</u> epimers.




D-Mannose epimer at C-2)

D-Glucose

D-Galactose (epimer at C-4)

# Carbohydrates

## Monosaccharides: The most important sugars

- Glyceraldehyde and dihydroxyacetone are the simplest (3 carbon) aldose and ketose, respectively.
- Ribose (Rib) is the standard five-carbon sugar.
- Glucose (Glc) is the standard six-carbon sugar.
- Galactose (Gal) is an C4-epimer of glucose.
- Mannose (Man) is an C2-epimer of glucose.
- Fructose (Fru) is the ketose form of glucose.
- Ribulose (Rul) is the ketose form of ribose.

Need to know, recognize, draw Fisher Projection, name, abbreviate

# Carbohydrates

## Monosaccharides: Stereoisomer Nomenclature

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  - Diastereomers have different physical properties (e.g., water solubility)
  - Example: ribose & lyxose
- Epimers
  - Epimers are stereoisomers that differ at only **one** chiral center
  - Epimers are diastereomers; diastereomers have different physical properties (i.e., water solubility, melting temp)
  - Example:
  - Glc & Man (2 epimer); Glc & Gal (4 epimer)
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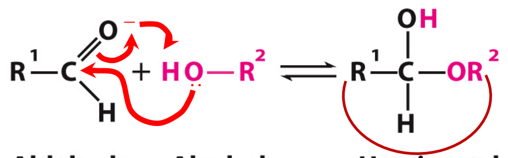
# Carbohydrates

## Monosaccharides: What are these “ring” forms?

- SUGARS WITH ≥5 CARBONS RAPIDLY AND STABLY FORM **RINGS** THROUGH HEMIACETAL (ALDOSES) AND HEMIKETAL (KETOSES) BONDS.

**Hemiacetals**



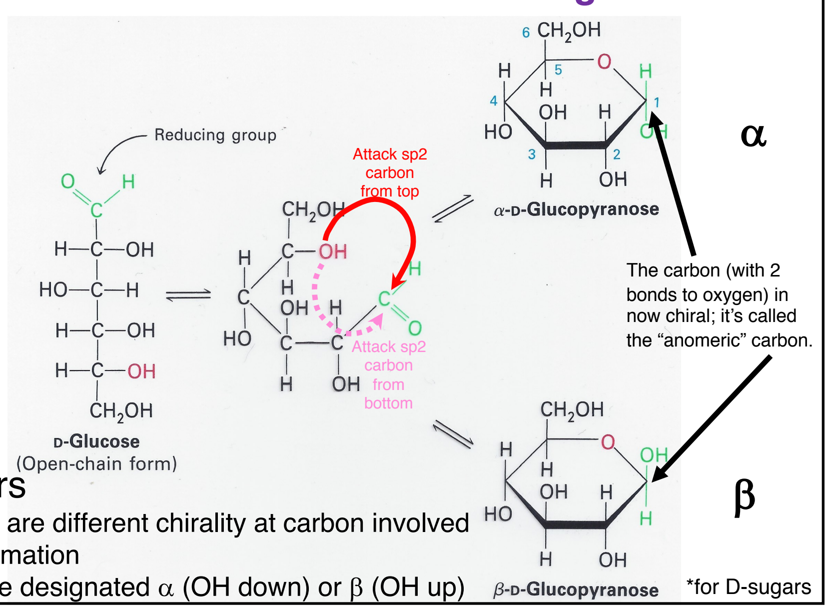
**Aldehyde** **Alcohol** **Hemiacetal**

Not chiral Chiral

- **IF** the aldehyde ($R^1$) and the alcohol ($R^2$) are on the same molecule, you have a RING!
- Due to the oxygen, the ring is heterocyclic

# Carbohydrates

## Monosaccharides: What are these “ring” forms?




Anomers

–Anomers are different chirality at carbon involved in ring formation

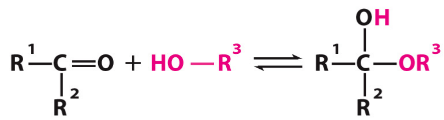
–These are designated α (OH down) or β (OH up)

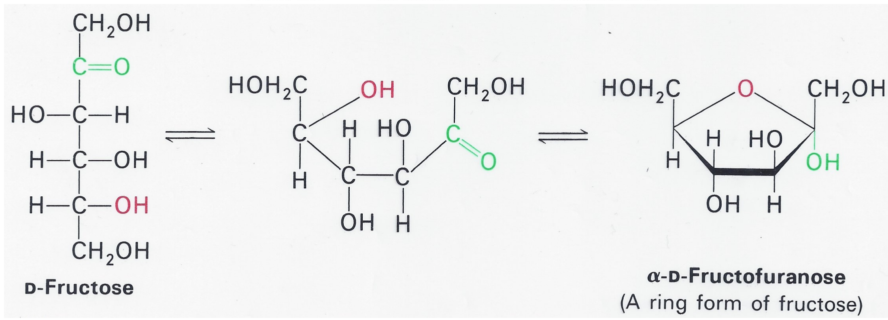
*for D-sugars

# Carbohydrates

## Monosaccharides: What are these "ring" forms?

**Hemiketals**




$\alpha$ (OH down) or $\beta$ (OH up)

Besides the $\alpha$ & $\beta$, what is this pyran and furan?

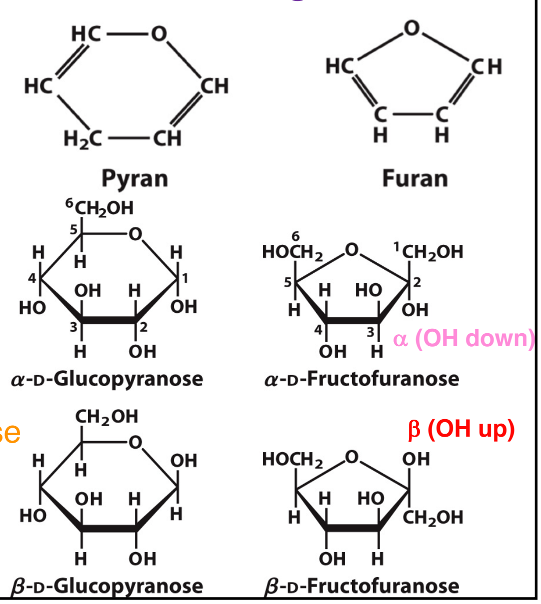
Its all about which alcohols are used in these reactions.

# Carbohydrates

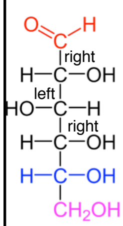
## Monosaccharides: What are these "ring" forms?

- Six-membered oxygen-containing rings are called pyranoses after the pyran ring structure.
- Five-membered oxygen-containing rings are called furanoses after the furan ring structure.

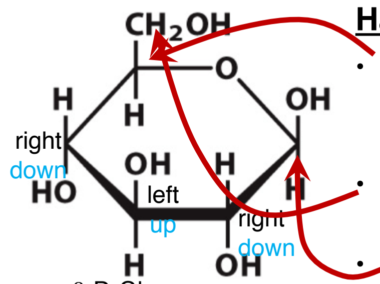
The way we are drawing these sugars is called a Haworth projection…..



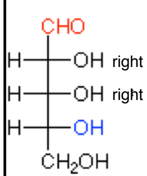
# Carbohydrates



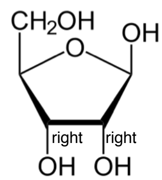

D-Glucose




β-D-Glucopyranose




D-ribose




β-D-Ribofuranose

## Haworth projections

- Highest carbon with hemiacetal/hemiketal alcohol in back; oxygen to right.
- If chiral (as for D-sugars), next highest carbon up if D, or down if L.
- The anomeric carbon is usually drawn on the right side.
- Other chiral carbons are down if right in Fisher projection and up if left in Fisher
- β-D-Glc pyranose goes up, down, up down, up as you go from C5 to C1
- β-D-Rib furanose goes up, down, down, up as you go from C4 to C1

# Carbohydrates

- Pentoses and hexoses readily undergo intramolecular cyclization.
- The former carbonyl carbon becomes a new chiral center, called the anomeric carbon.
- ALL forms are in equilibrium in solution

  Process is called "mutarotation"

Furan

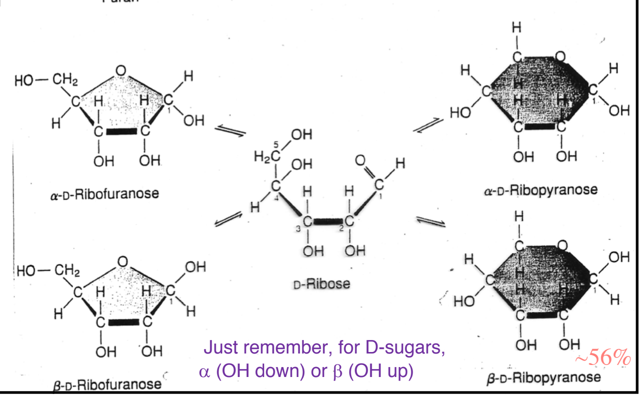
Ring Forms (Anomers)

Pyran

– When the former carbonyl oxygen becomes a hydroxyl group, the position of this group determines if the anomer is $\alpha$ or $\beta$.

– If the hydroxyl group is on the opposite side (trans) of the ring as the $CH_2OH$ moiety, the configuration is $\alpha$.

– If the hydroxyl group is on the same side (cis) of the ring as the $CH_2OH$ moiety, the configuration is $\beta$.




Just remember, for D-sugars, α (OH down) or β (OH up)