

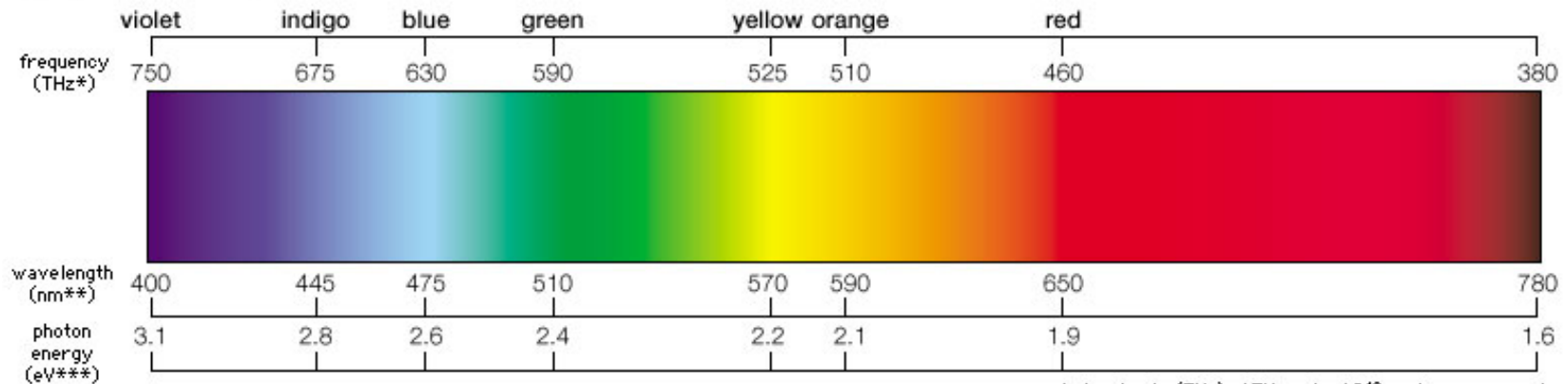
Introduction to Photonics

REU in Photonics – Summer 2012

Boston University

Welcome !!

Light, the visible spectrum

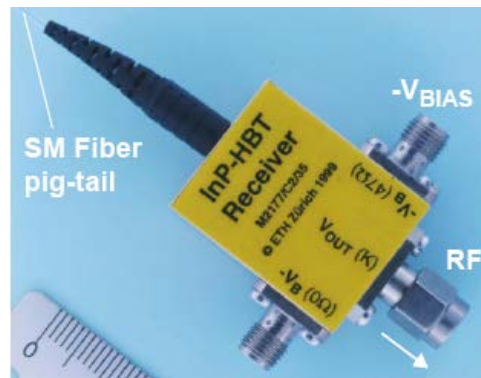


* In terahertz (THz); 1THz = 1×10^{12} cycles per second.
** In nanometres (nm); 1nm = 1×10^{-9} metre.
*** In electron volts (eV).

Where do we use optics/photronics?

Optical Engineers work in:

- Optical communications
- Information and computing
- Medical imaging
- Biology and bio-chemical sensing
- Defense applications
- Navigation and control systems
- Image processing
- Data storage / memory



PHOTONICS/OPTICS

1- What is light?

2- How can we generate light?

3- How does light interact with materials or itself?

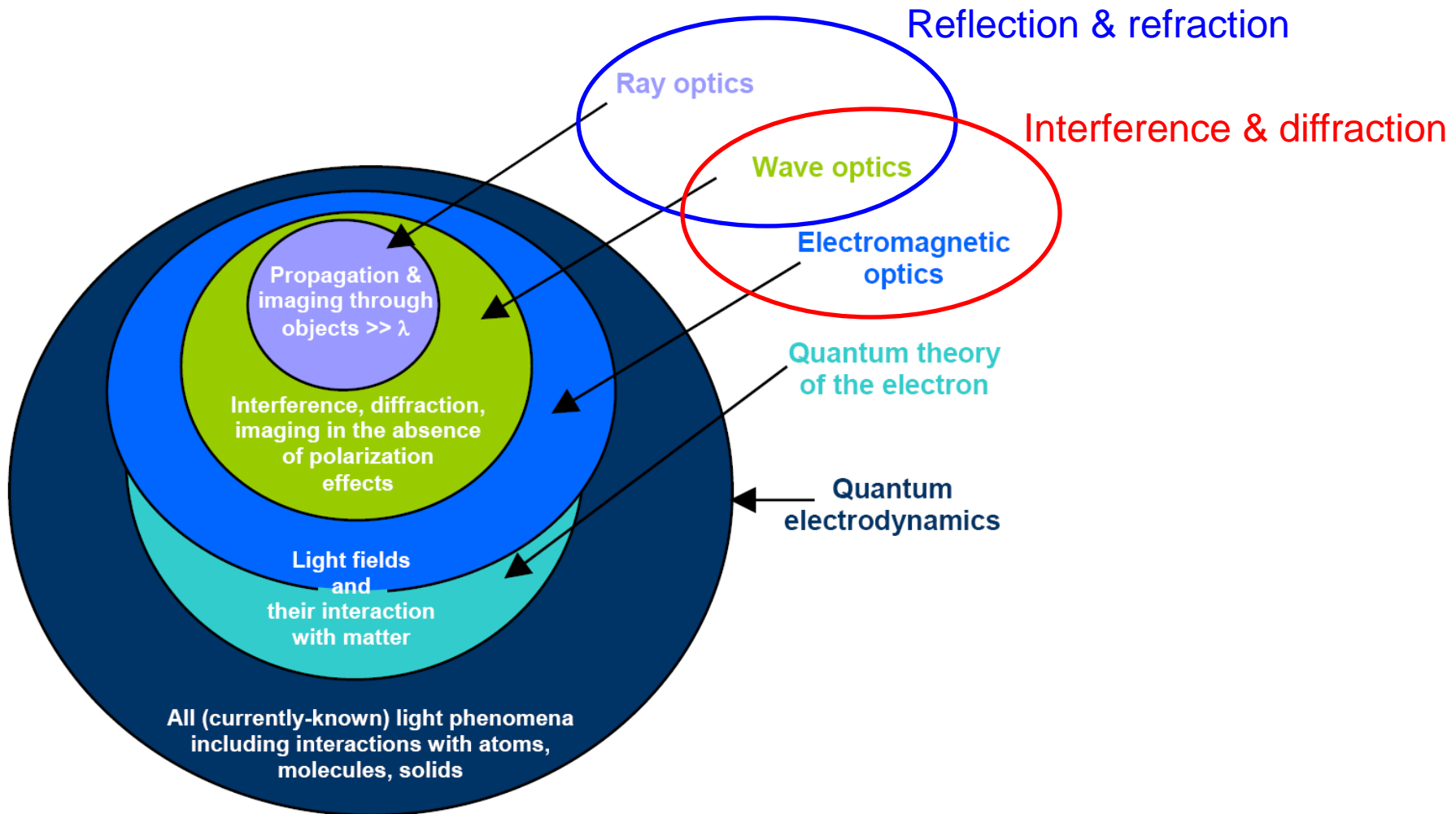
4- How can we manipulate light?

5- How can we measure light?

6- What can we do with light?

THEORIES of LIGHT

Understanding the nature of light and its interaction with materials



Optical waves

Light consists of coupled electric and magnetic fields that satisfy a wave equation:

$$\nabla^2 \vec{E} - \frac{1}{c^2} \frac{\partial^2 \vec{E}}{\partial t^2} = 0$$

c : speed of light (3×10^8 m/sec in free space)

which has simple sine-wave solutions: $\vec{E}(\vec{r}, t) \propto \cos(\omega t - kz)$

DEFINITIONS:

$\omega = 2\pi\nu$: angular frequency [rad/sec]

$\nu = 1/T$: frequency [Hz]

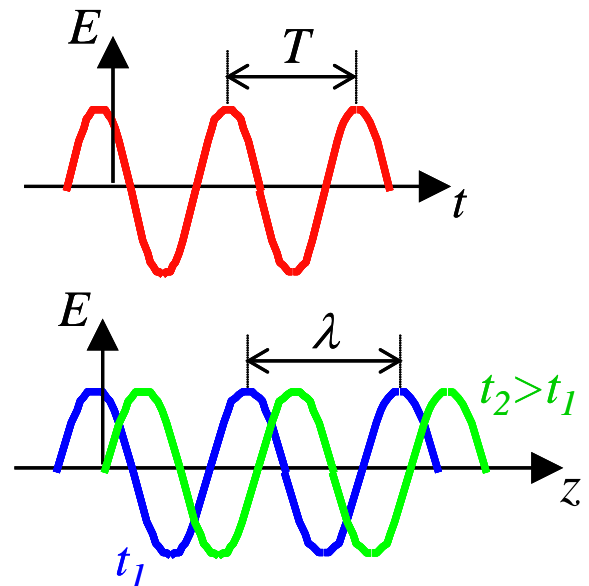
T : optical period [sec]

$k = 2\pi/\lambda$: wavenumber [m^{-1}]

λ : wavelength [m]

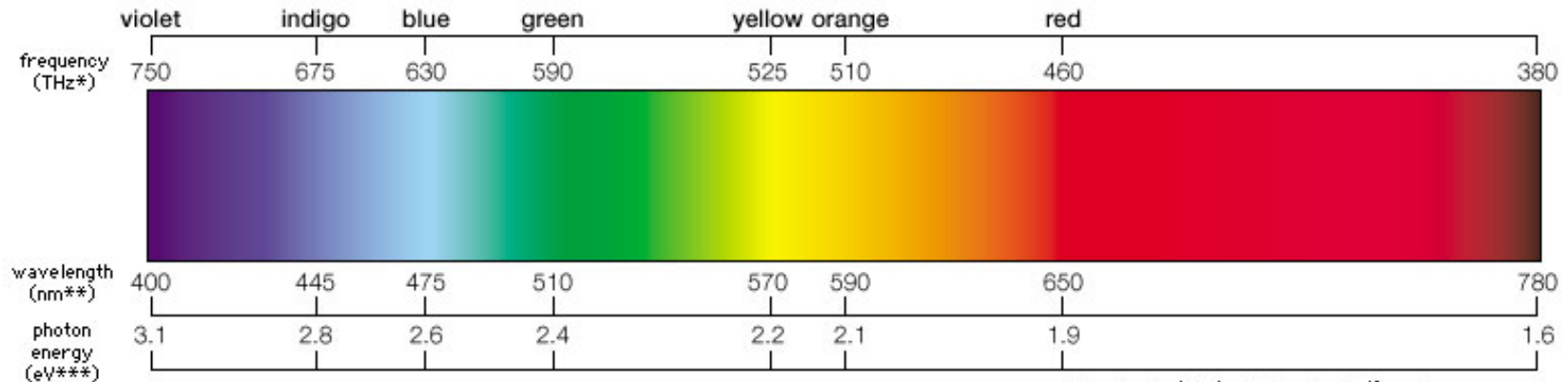
From the wave equation: $\omega = ck \Leftrightarrow \nu = c/\lambda$

HARMONIC PLANE WAVES:



Wavelength and color

Light, the visible spectrum



* In terahertz (THz); 1THz = 1×10^{12} cycles per second.
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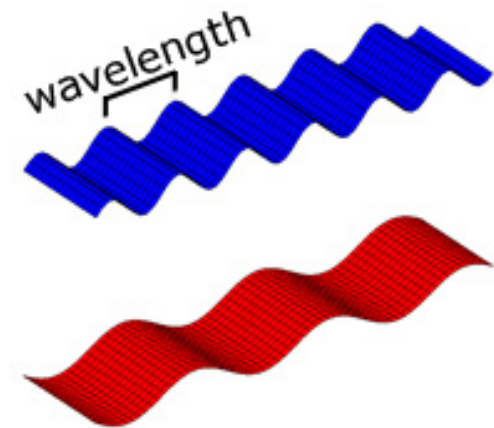
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MAIN SPECTRAL REGIONS:

$10 \text{ nm} < \lambda < 400 \text{ nm}$: ultraviolet

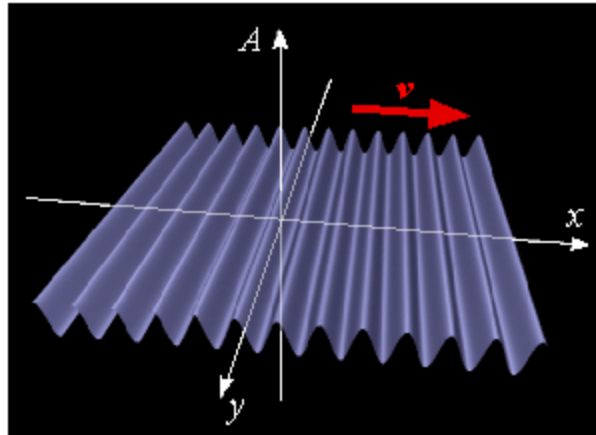
$400 \text{ nm} < \lambda < 780 \text{ nm}$: visible

$780 \text{ nm} < \lambda < 1 \text{ mm}$: infrared



Optical waves

SIMPLE
TYPES:



Plane wave



Spherical wave

OTHER KEY PROPERTIES:

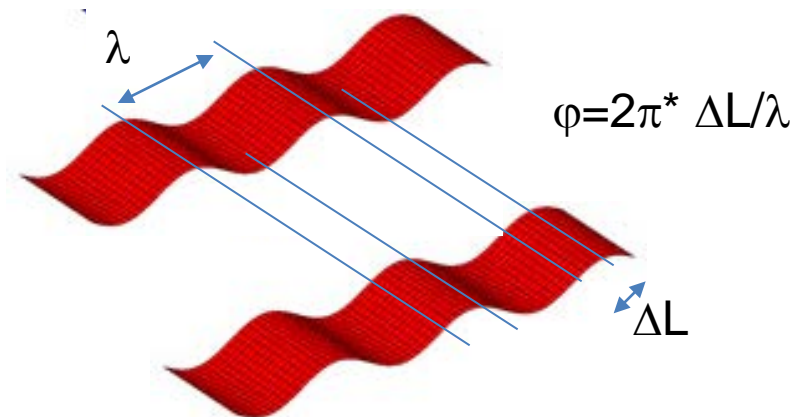
Intensity I

I : energy carried per unit area per unit time (proportional to the electric field amplitude squared)

$I = \phi h\nu$, where h is Planck's constant and ϕ is an integer (photon flux)

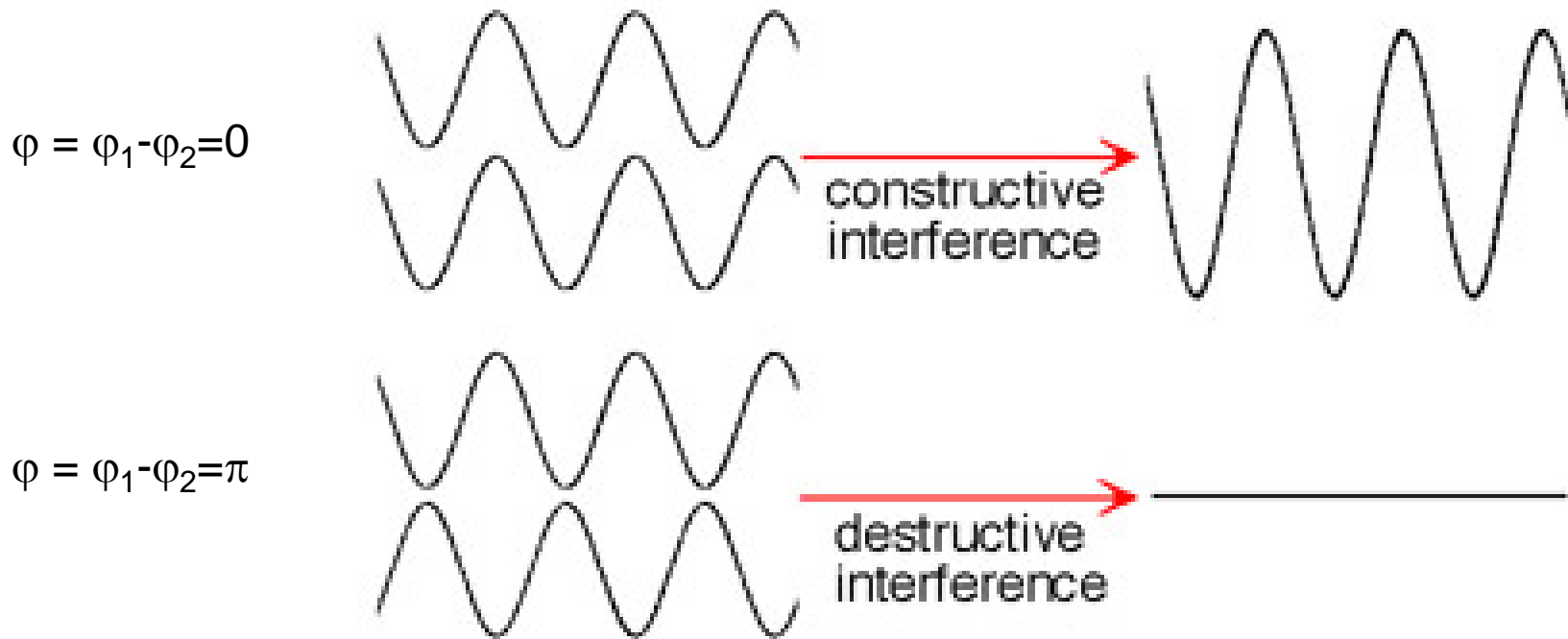
PHOTON: "particle" of light

Phase ϕ (determines interference)



In phase when $\Delta L = n\lambda$ ($n=0, \pm 1, \pm 2, \dots$)

Interference conditions



Constructive Interference:

→ IN-PHASE

Phase Difference is $0, 2\pi, 4\pi, \dots$

Path Difference is $\lambda, 2\lambda, 3\lambda, \dots$

Destructive Interference:

→ OUT-of-PHASE

Phase Difference is $1\pi, 3\pi, 5\pi, \dots$

Path Difference is $1\lambda/2, 3\lambda/2, 5\lambda/2, \dots$

Interference of two waves

Intensity of wave 1: $I_1 = a_1^2$

Intensity of wave 2: $I_2 = a_2^2$

Phase difference: $\varphi = \varphi_1 - \varphi_2$

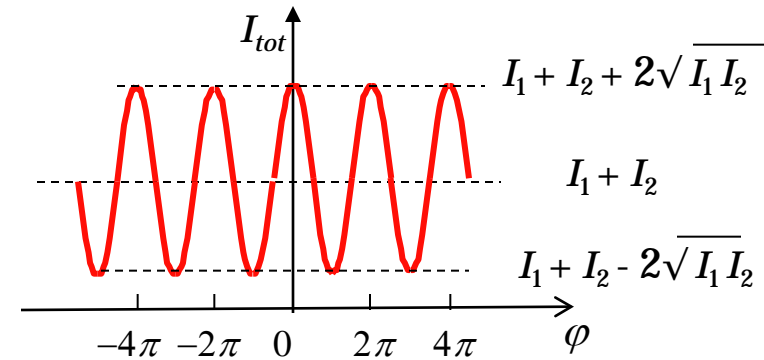
$$I_{tot} = I_1 + I_2 + 2(I_1 I_2)^{1/2} \cos \varphi$$

INTERFERENCE EQUATION

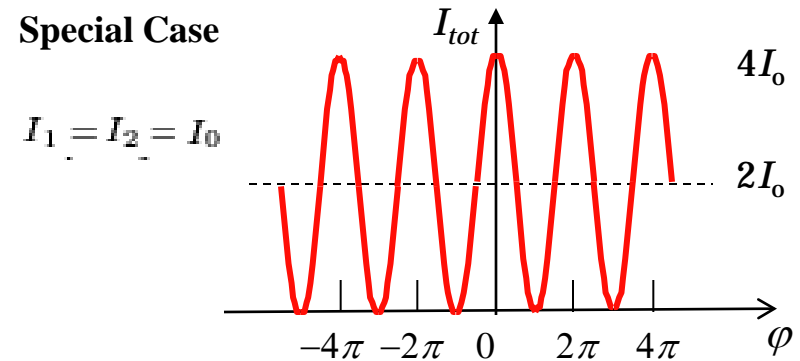
Max Intensity → **Constructive Interference**
Occurs when waves are **IN-PHASE**

Min Intensity → **Destructive Interference**
Occurs when waves are **OUT-PHASE**

Intensity of superposition depends on the intensities of the individual waves and on their phase difference



Special Case

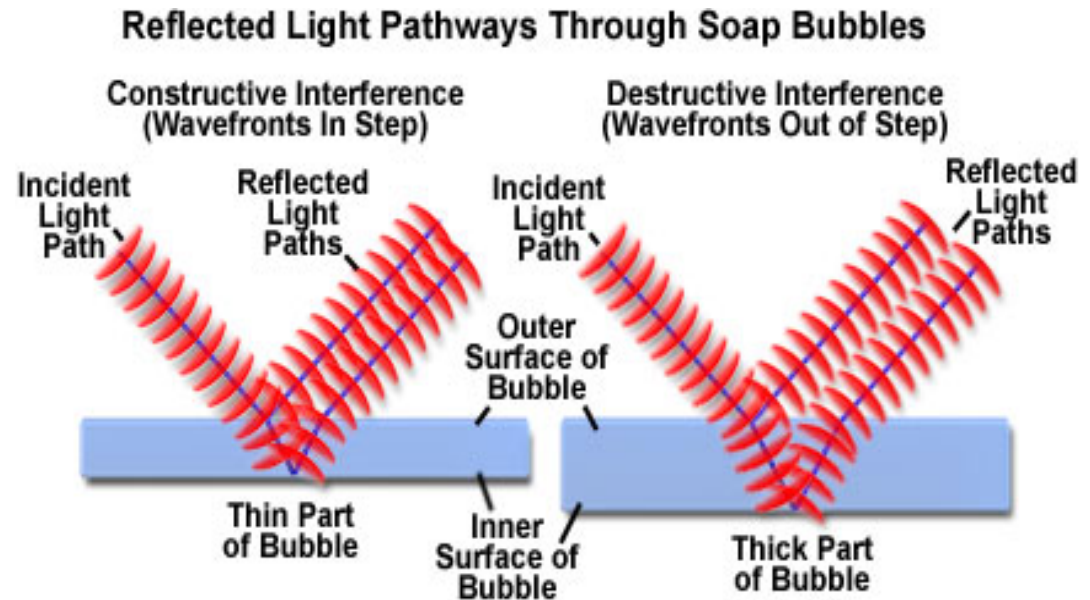


Thin film interference: science of soap bubbles

Soap Film Interference Patterns



Thickness variations and angle of incident light changes color



The phase difference between the two reflected waves depends on the difference in their path length and on the wavelength

Color via interference



Soap Film Interference Patterns



Figure 1



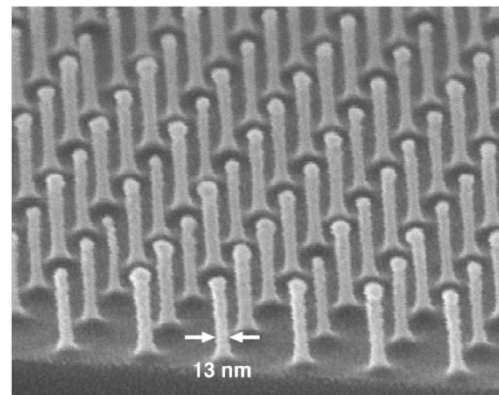
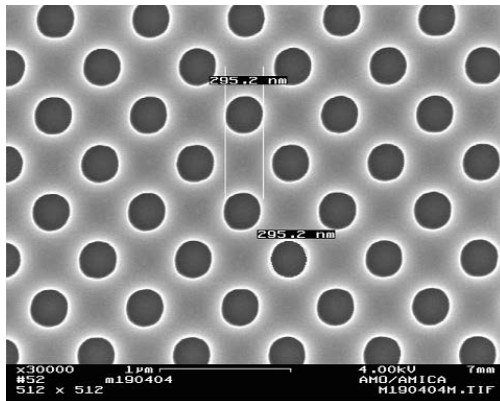
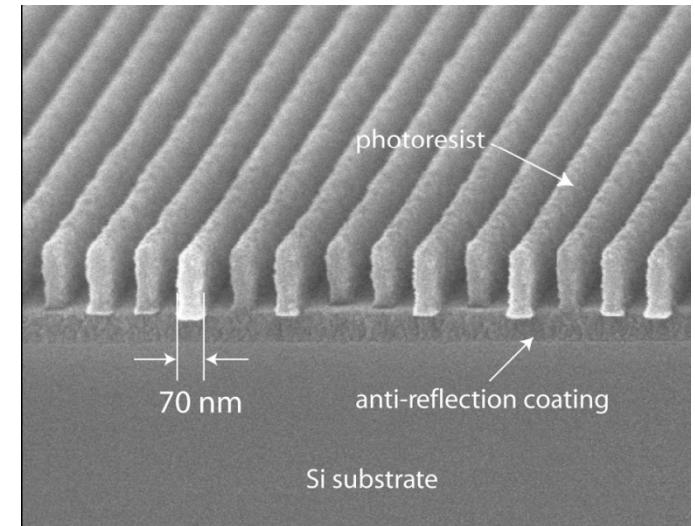
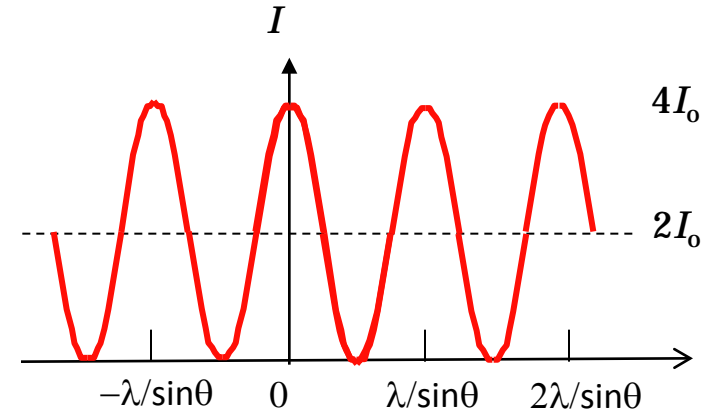
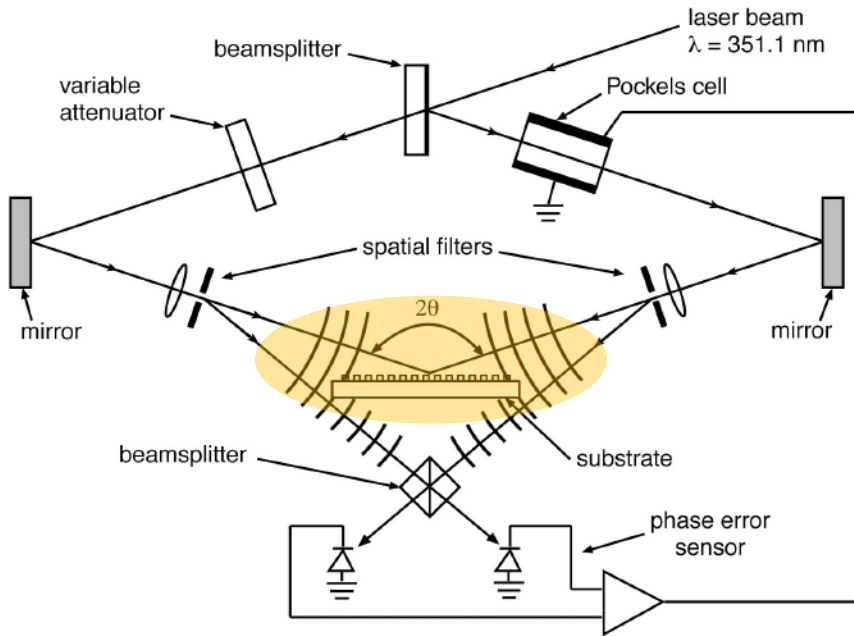
Interference based displays

Anti-reflection coating

Interference filters



Interference lithography

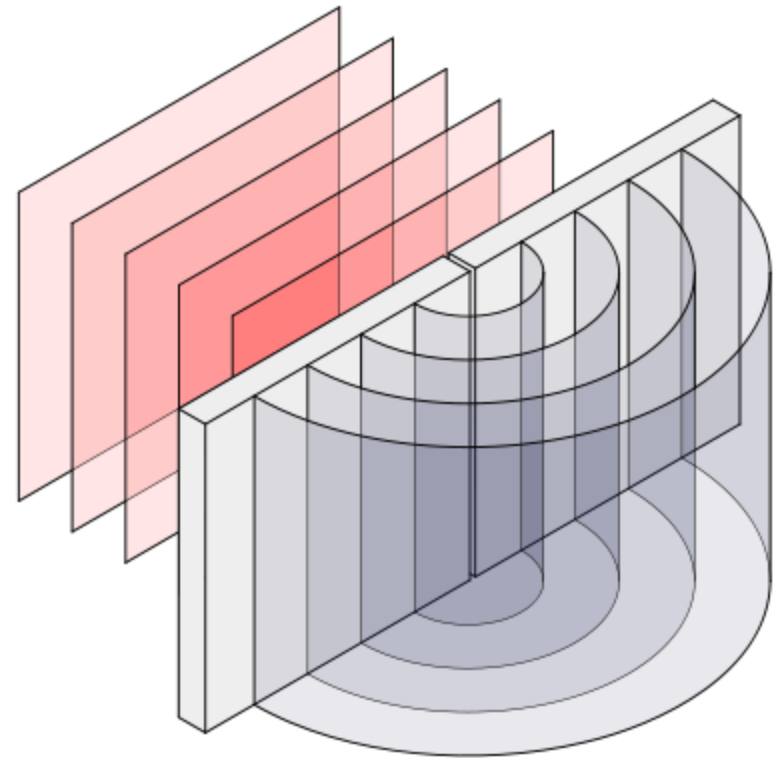


100nm-period posts in Si

Single exposure → lines
Multiple exposure → 2D patterns

Diffraction

Diffraction: “ A modification which light undergoes especially in passing by the edges of opaque bodies or through narrow openings and in which the rays appear to be deflected”



Diffraction of water waves

Diffraction occurs for all waves



Ocean waves passing through slits in Tel Aviv, Israel

Diffraction: wave phenomenon

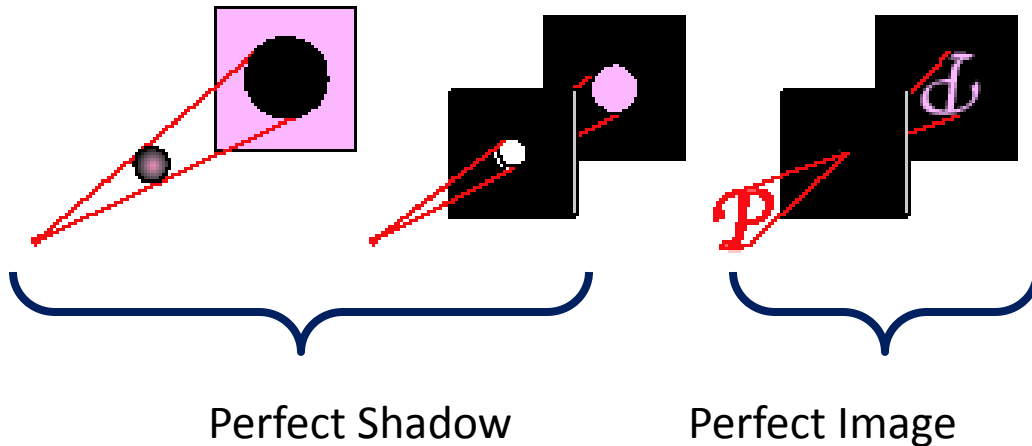
Diffraction: **Wave phenomenon**

Occurs whenever a portion of a wavefront (light, sound, water) is obscured

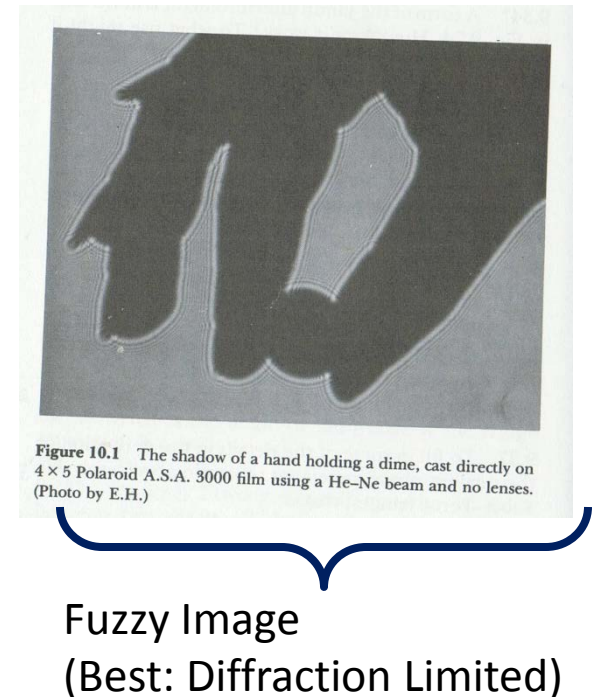
- * wave encounters the edges of an opaque body
- * wave encounters an aperture in an opaque film

Deviation of light from rectilinear propagation

RAY OPTICS



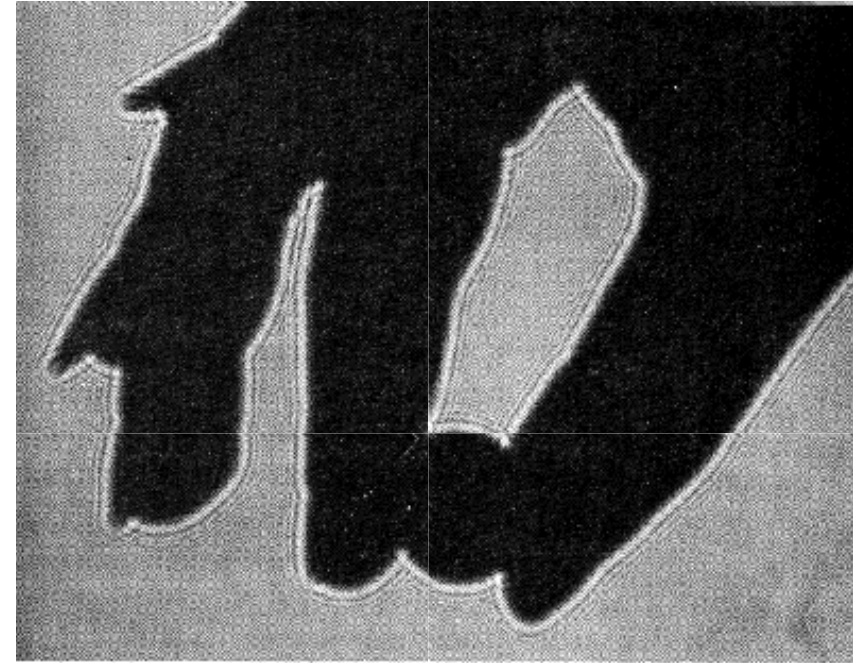
WAVE OPTICS



Diffraction: wave phenomenon

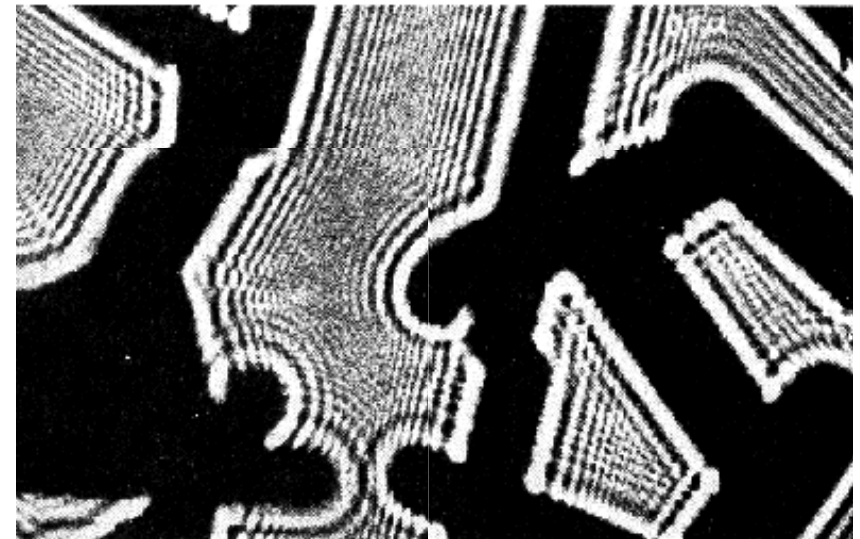
1) Light does not always travel in a straight line.

Shadow of a hand illuminated by a Helium-Neon laser



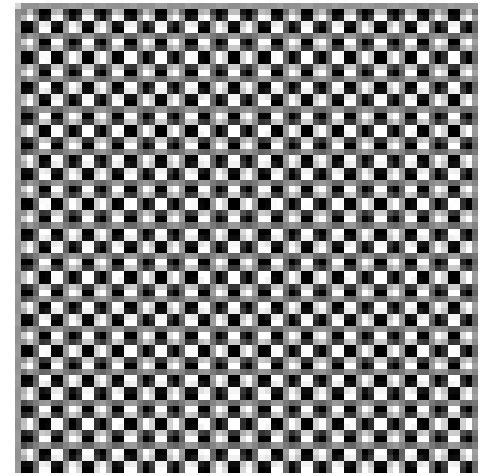
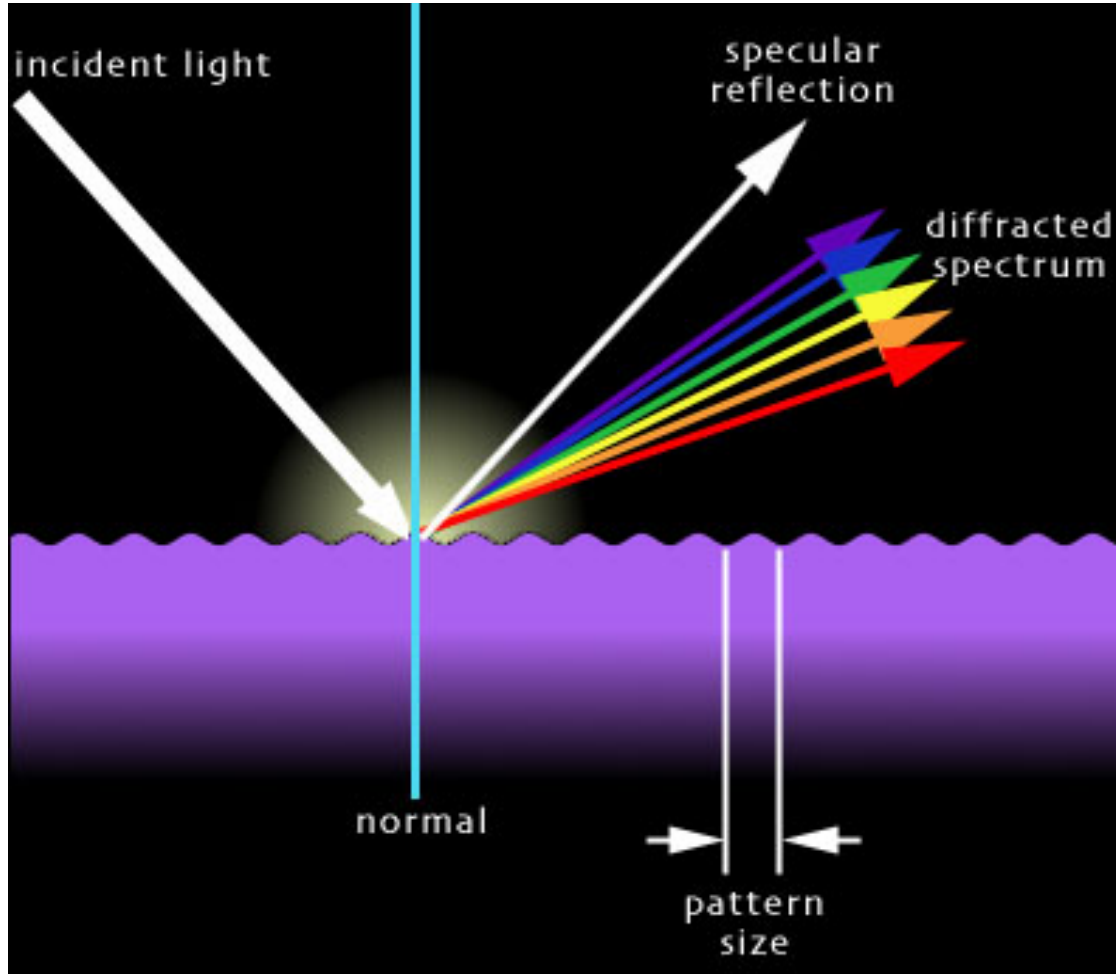
2) It tends to bend around objects. This tendency is called **diffraction**.

Shadow of a zinc oxide crystal illuminated by a electrons

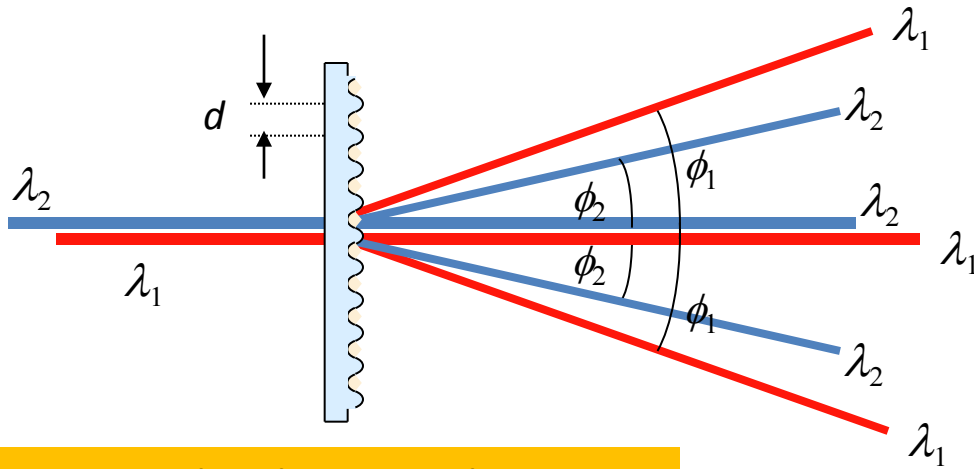


3) Any wave will do this, including matter waves and acoustic waves.

Diffraction gratings



Grating spectrometers



$$d \sin(\phi) = m\lambda$$
$$m=0, \pm 1, \pm 2, \dots$$

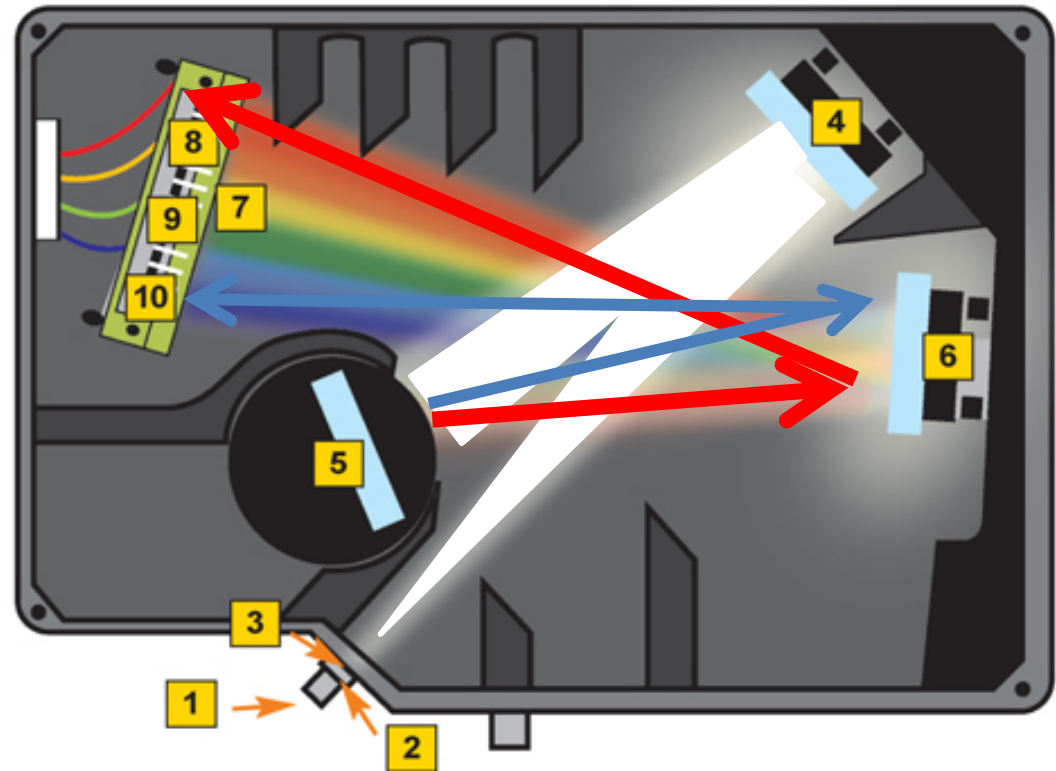
$$\text{for } \lambda \ll d \quad \left\{ \begin{array}{l} \phi_1 \approx \lambda_1/d \\ \phi_2 \approx \lambda_2/d \end{array} \right.$$

We can make objects such as gasses glow by heating them up in a flame, or by passing electricity through them.

Different elements emit light with different colors when they glow.

Spectroscopy works by taking light and splitting it up into its component colors.

→ we can identify the elements by the bright lines we see in the spectroscope.



Light sources

SEVERAL DIFFERENT LIGHT EMISSION MECHANISMS:

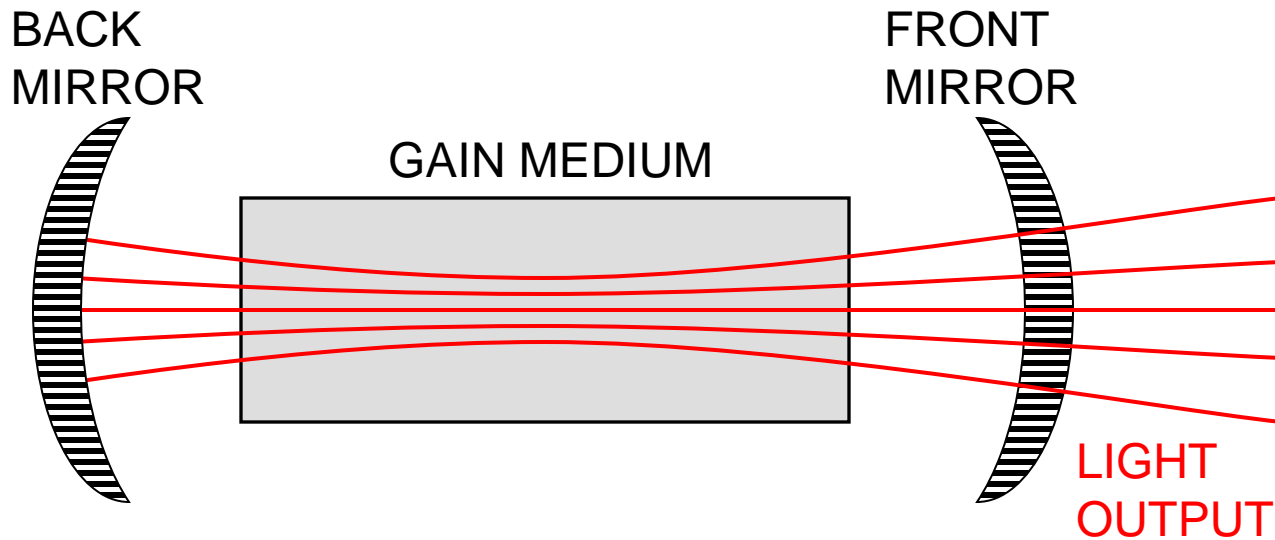
INCANDESCENCE: light emission from a hot medium due to its temperature [Examples: the sun, incandescent light bulbs]

LUMINESCENCE: light emission from a medium in an excited energy state when it relaxes to its ground state spontaneously [Examples: phosphors, LEDs]

STIMULATED EMISSION: light emission from a medium in an excited energy state when it relaxes to its ground state under the action of an electromagnetic wave [lasers]

Laser: Light Amplifier by Stimulated Emission of Radiation

Lasers

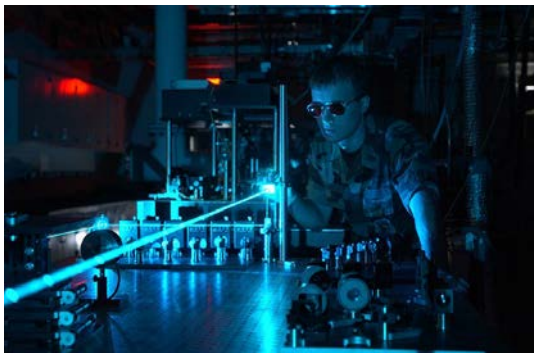
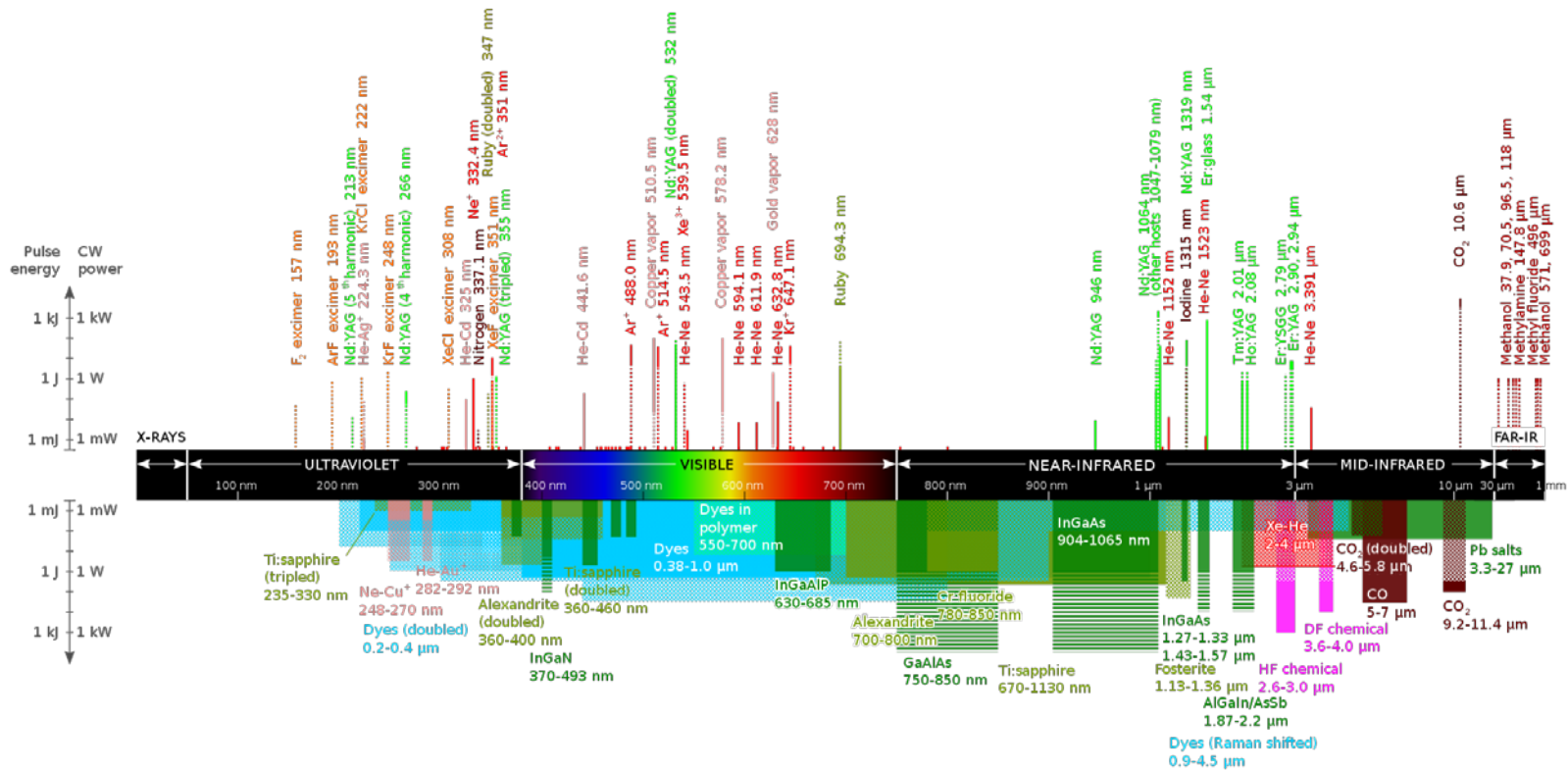


TWO MAIN INGREDIENTS:

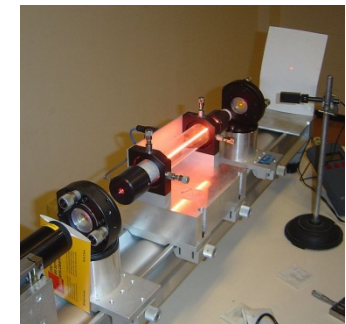
GAIN MEDIUM: a sample of some material capable of amplifying light at certain wavelengths (when suitably pumped)

OPTICAL CAVITY: two mirrors aligned about a common axis, providing optical feedback to the gain medium

Available laser wavelengths



Diode laser

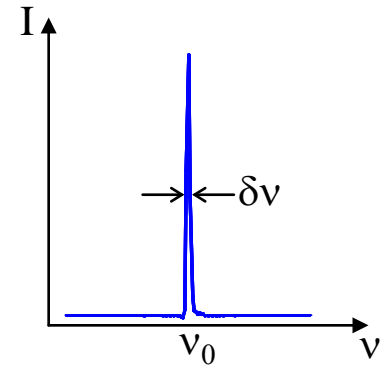


HeNe laser

Laser light

KEY PROPERTIES (for a HeNe laser as an example):

Quasi-monochromatic: very narrow spectrum with $\delta\nu/\nu$ as small as 10^{-10} (versus 10^{-1} for thermal light)



Highly directional: divergence angle as small as 10^{-4} rad

Highly coherent: phase remains constant over very long propagation distances (a few 10 km, versus μm for thermal light)

High peak intensity: over 12 orders of magnitude higher than sunlight (per unit bandwidth, per unit solid angle)