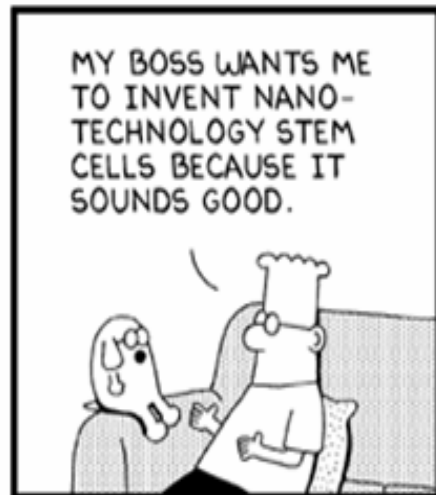




Stronger than Steel: Carbon Nanotubes

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Jessica Kaufman
Marc McGuigan
Boston University
December 15, 2005

How small is nano?



Nano-hype!

Today's Nanotube Workshop

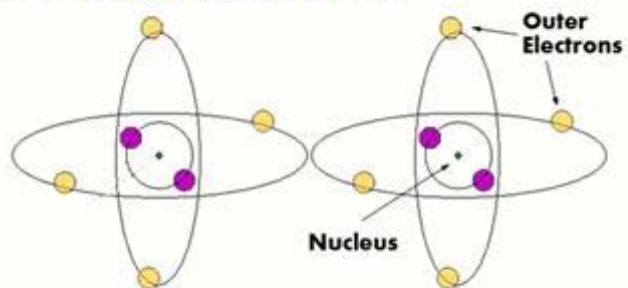


- The Chemistry of Carbon
 - A New Form of Carbon: Nanotubes
 - Activity: Building Nanotube Models
- Nanotubes in Biology and Medicine
 - Biosensors
 - Filtration of Biocontaminants
- Carbon Nanotubes – Stronger than steel
 - Tensile strength of Carbon Nanotubes
 - Composite Materials
 - Applications

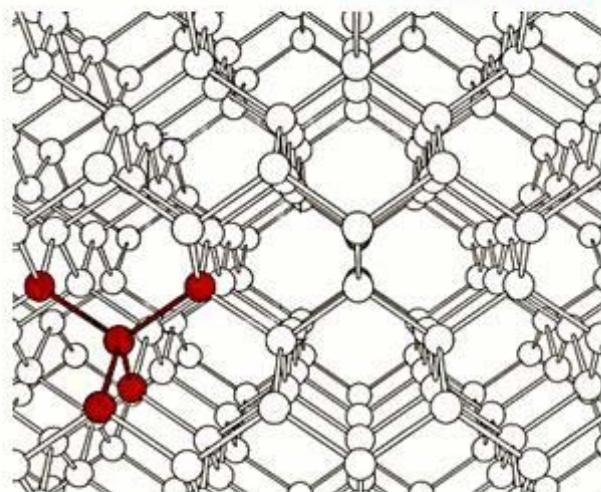
Chemistry of Carbon



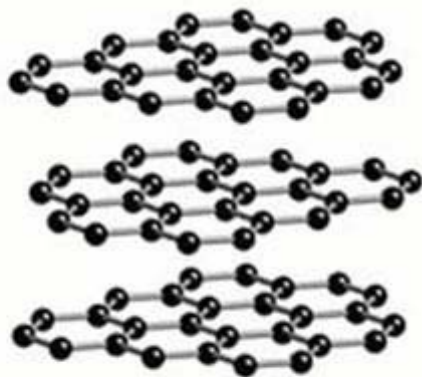
Carbon Atoms



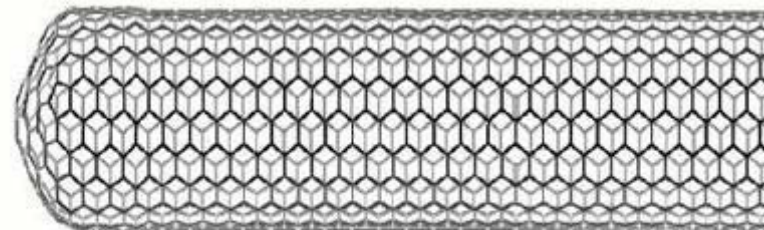
Diamond



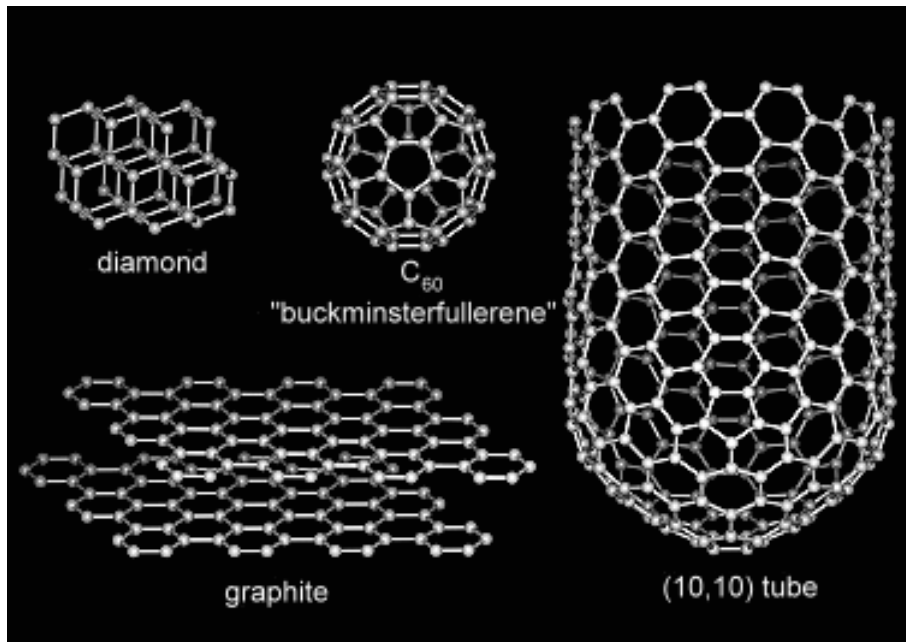
Graphite



Nanotube

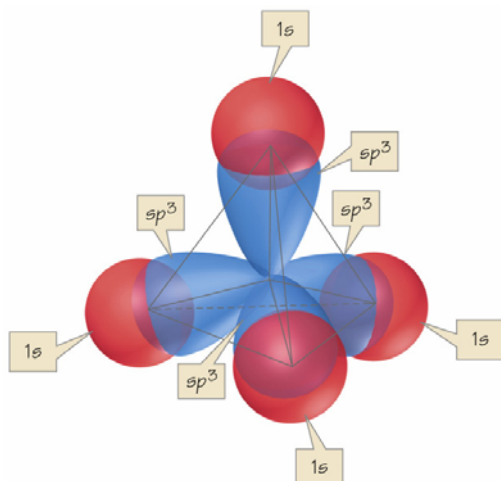


Forms of Carbon



- Carbon was discovered in prehistory and was known to the ancients
- The allotropes of carbon are the different molecular configurations that pure carbon can take.
- The three relatively well-known allotropes of carbon are amorphous carbon, graphite and diamond.
- Several exotic allotropes have also been synthesized or discovered, including fullerenes and carbon nanotubes.

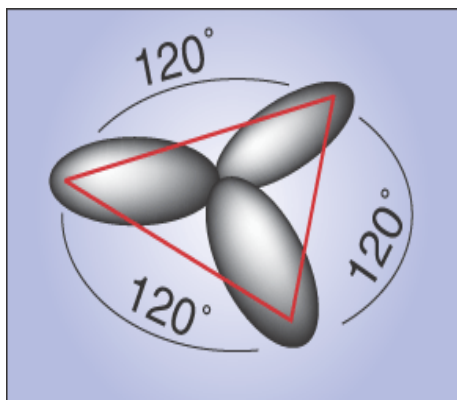
Graphite



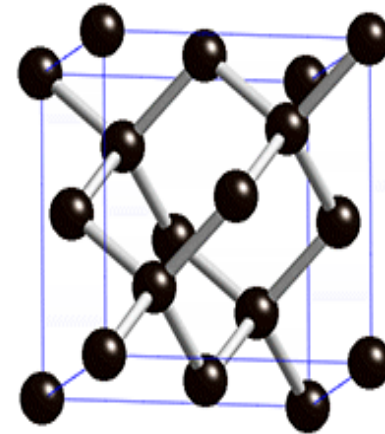
- Graphite is a conductor, and can be used as the material in the electrodes of an electrical arc lamp, or as the lead in pencils.

- Each carbon atom is covalently bonded to three other surrounding carbon atoms.

- Each carbon atom possesses an sp^2 orbital hybridization. The pi orbital electrons delocalized across the hexagonal atomic sheets of carbon contribute the graphite's conductivity.

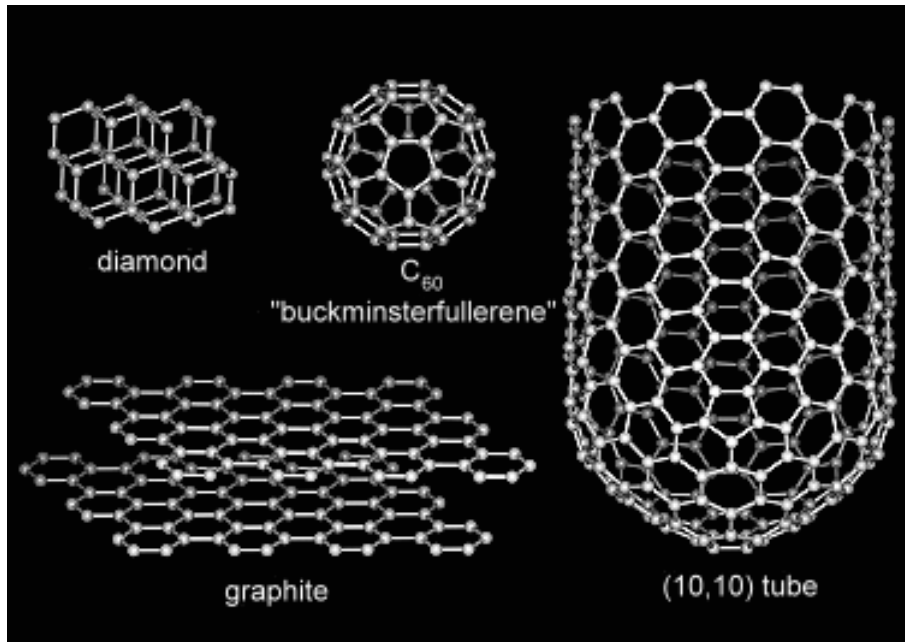


Diamonds



- Diamonds typically crystallize in the cubic crystal system and consist of tetrahedrally bonded carbon atoms
- The tetrahedral arrangement of atoms in a diamond crystal is the source of many of diamond's properties; graphite, has a rhombohedral crystal structure and as a result shows dramatically different physical characteristics—contrary to diamond, graphite is a very soft, dark grey, opaque mineral.

History of Carbon Nanotubes

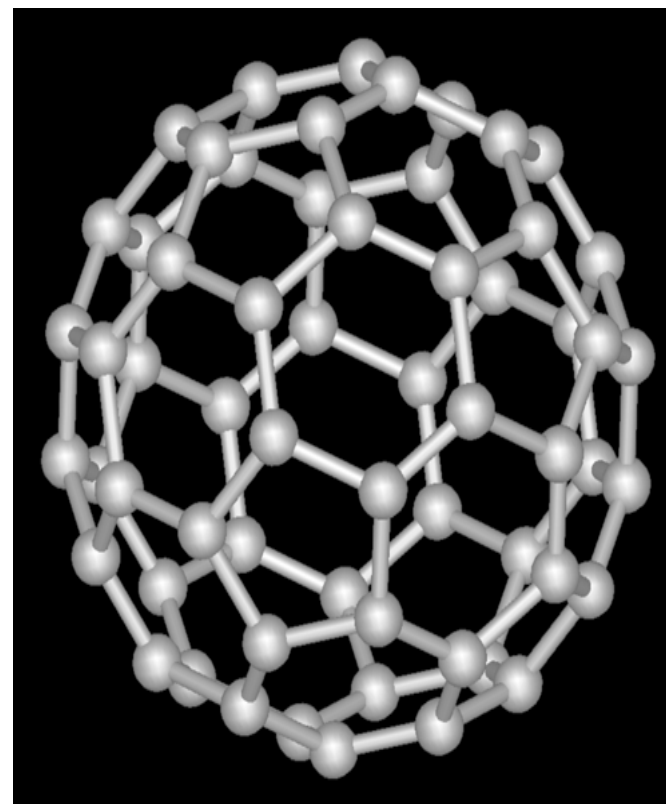


- In 1985 a new class of molecules consisting of purely carbon was discovered, the Fullerenes.
- In 1990 Dr. Richard Smalley recognized that tubular Fullerenes should be possible.
- It was not until 1991 that Sumio Iijima of the NEC Laboratory in Tsukuba, Japan, observed that these fibers were hollow.

Fullerenes



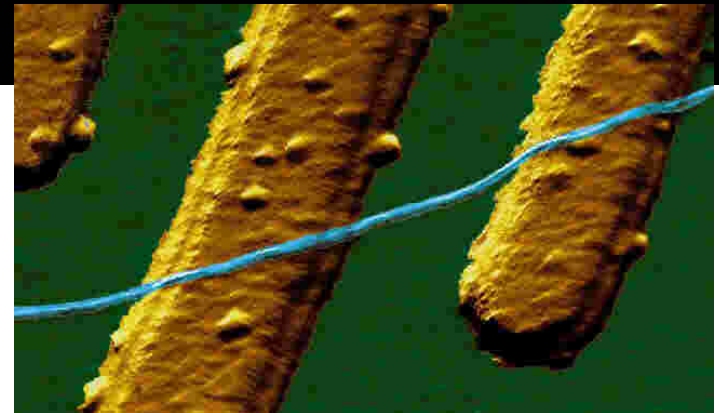
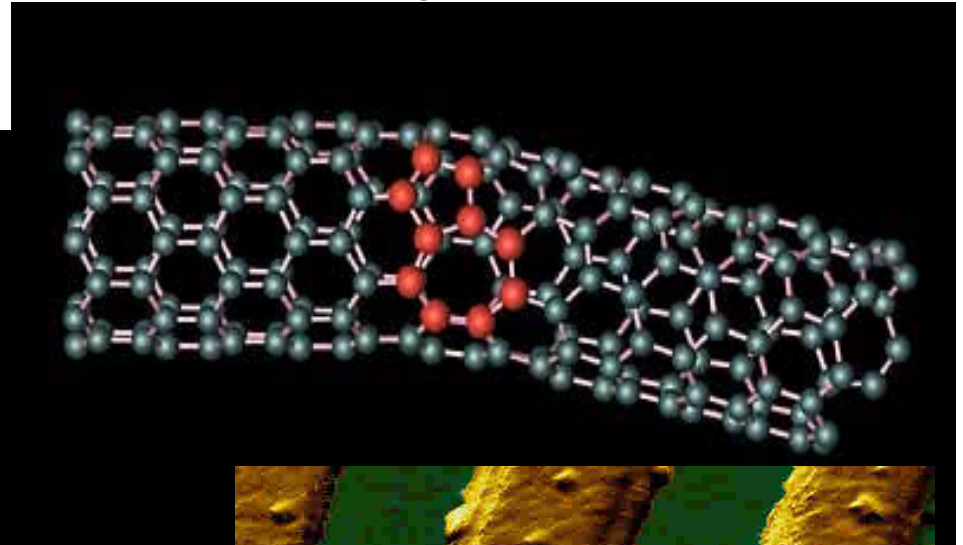
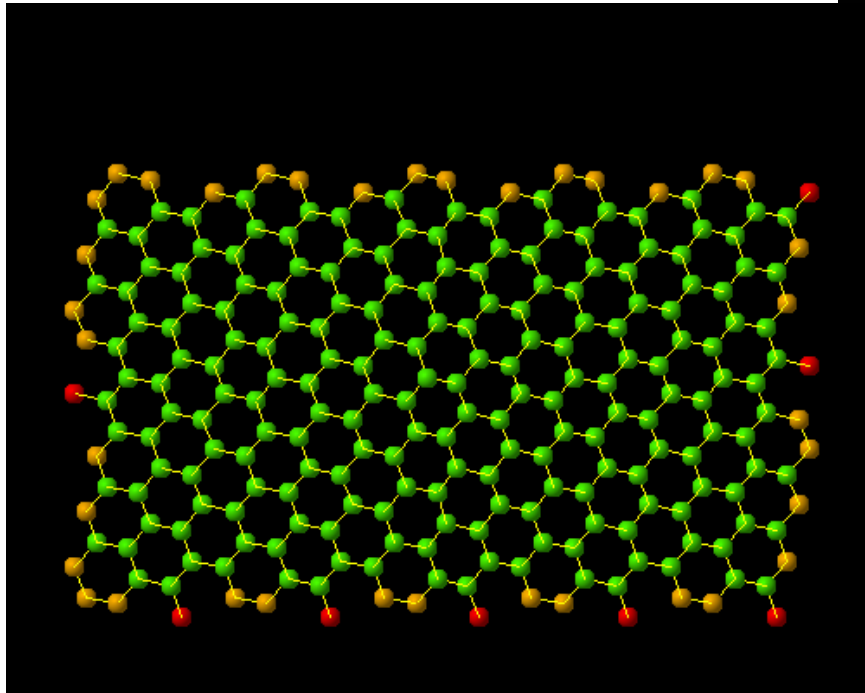
- The Fullerenes are a recently discovered allotrope of carbon.
- Fullerenes are not very reactive due to the stability of the graphite-like bonds



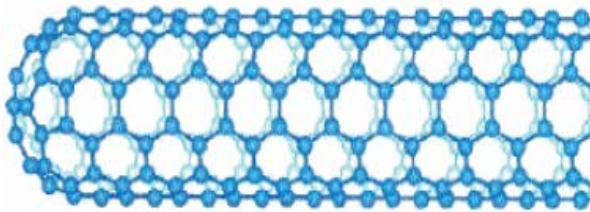
Carbon Nanotubes



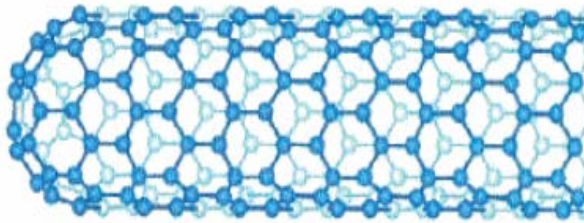
- Carbon nanotubes are cylindrical carbon molecules with novel properties that make them potentially useful in a wide variety of applications.
- Nanotubes are composed entirely of sp^2 bonds, similar to graphite.



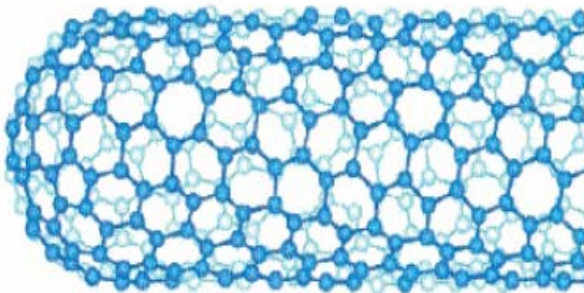
What makes nanotubes different from one another?



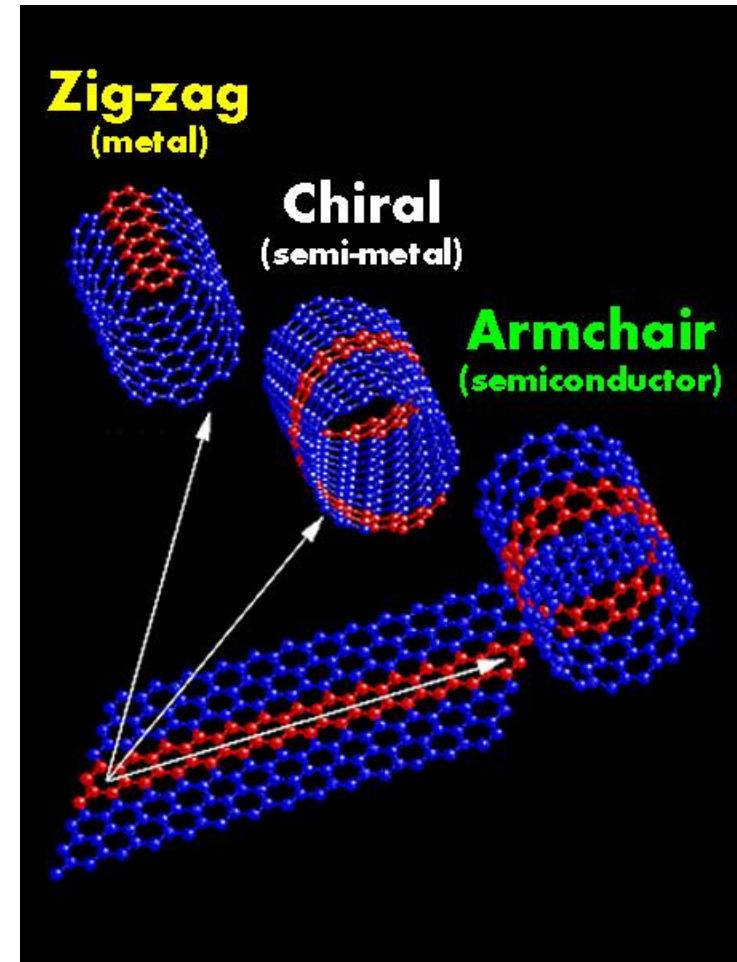
Armchair ($\alpha = 30^\circ$)



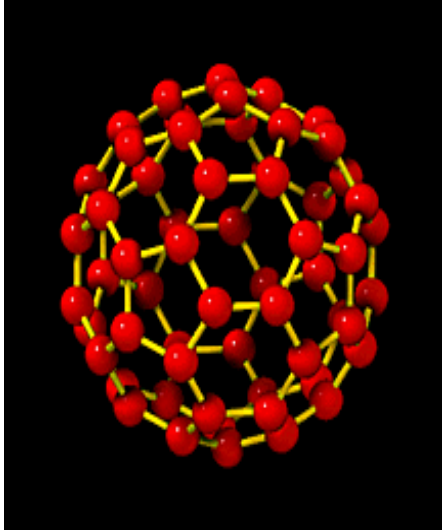
Zigzag ($\alpha = 0^\circ$)



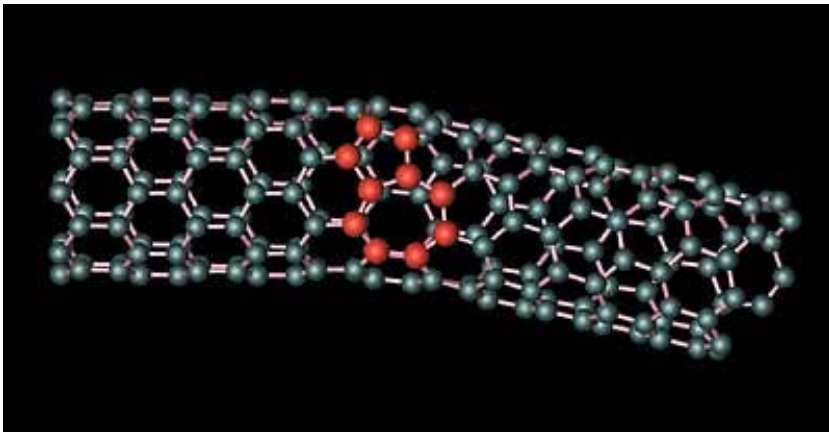
Intermediate ($0 < \alpha < 30^\circ$)



Inquiry Activity



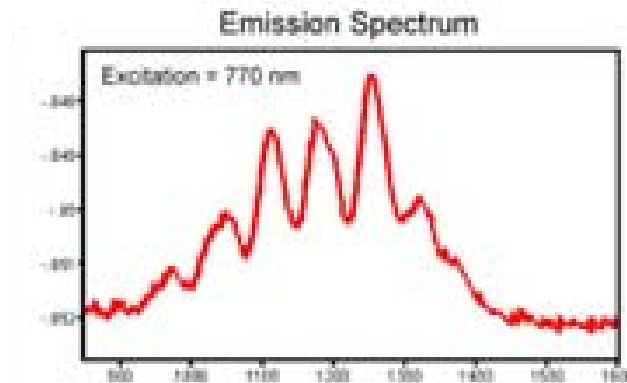
Buckyballs



Nanotubes

Ideal sensor molecule

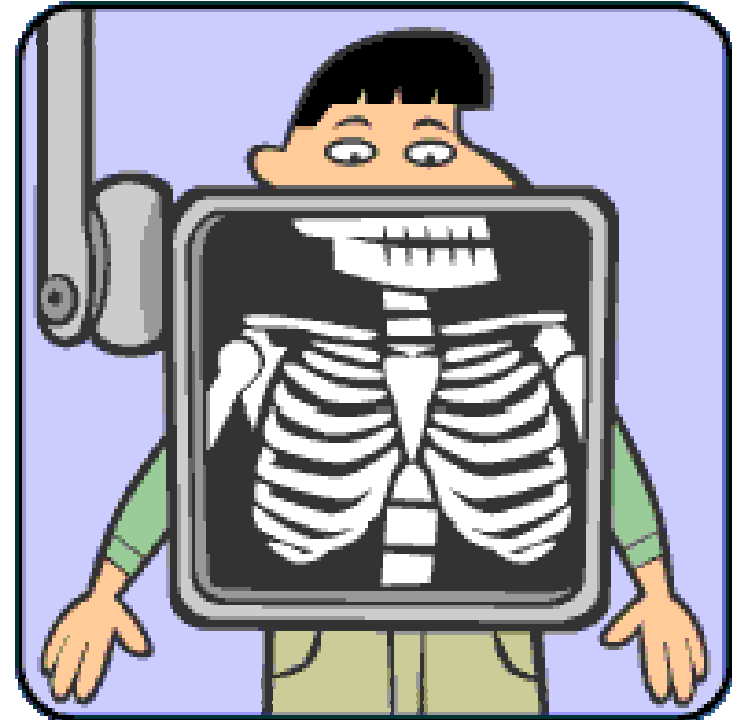
- Remarkable Properties of CNTs
 - Carbon Nanotubes “glow”
 - Excited in the near-IR range of the EM spectrum
 - Different than almost all other dyes
 - Continuous fluorescence- no photobleaching
- Ideal for sensor applications in the body



Nanotubes and medicine



- Fluorescence of nanotubes can be used to make biosensors
- Light and the human body
 - Different wavelengths interact with tissue differently
 - Familiar with X-rays



<http://pbskids.org/lions/words/xray.html>



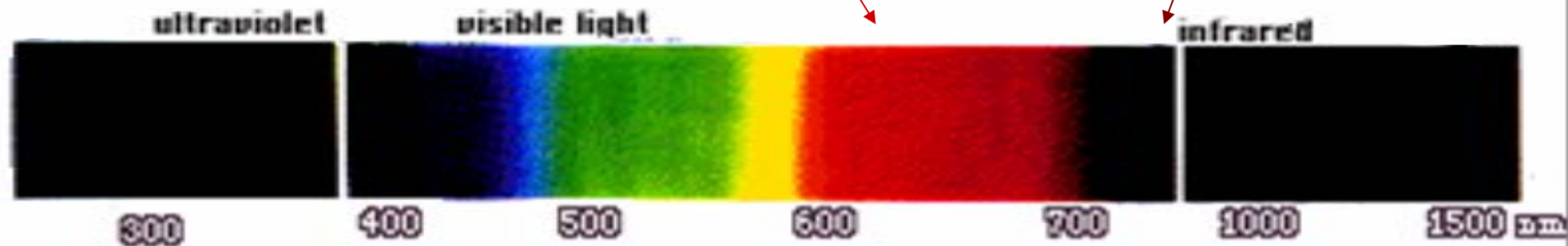
Window into tissue

“There is a window in which light can penetrate into tissue: the near-IR wavelengths, from approximately 700 to 900 nm.”

(<http://www.spie.org/web/oer/may/may00/cover2.html>)

Red is good

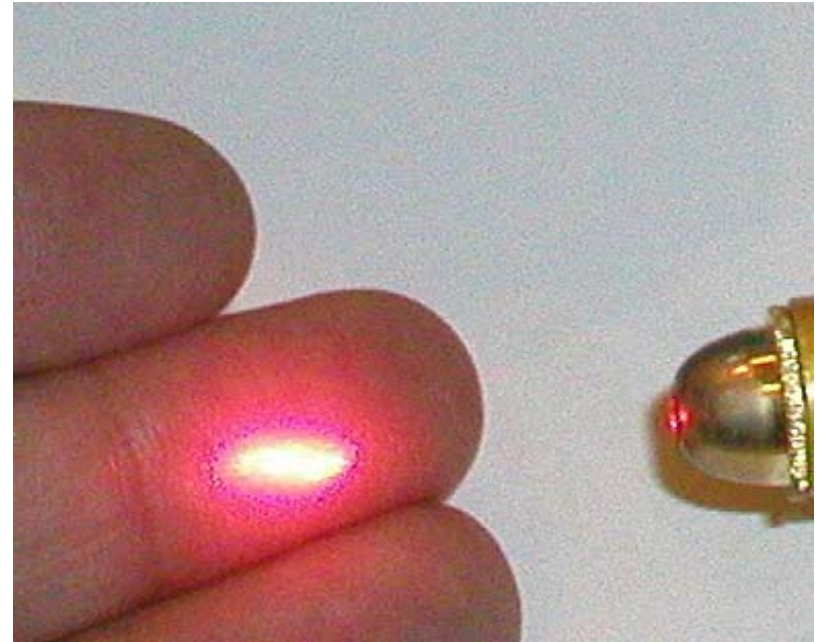
Near-IR is best



Lesson engagement



- How would you choose to test your blood?



From <http://www.vitrex.dk/images/lancets.jpg>

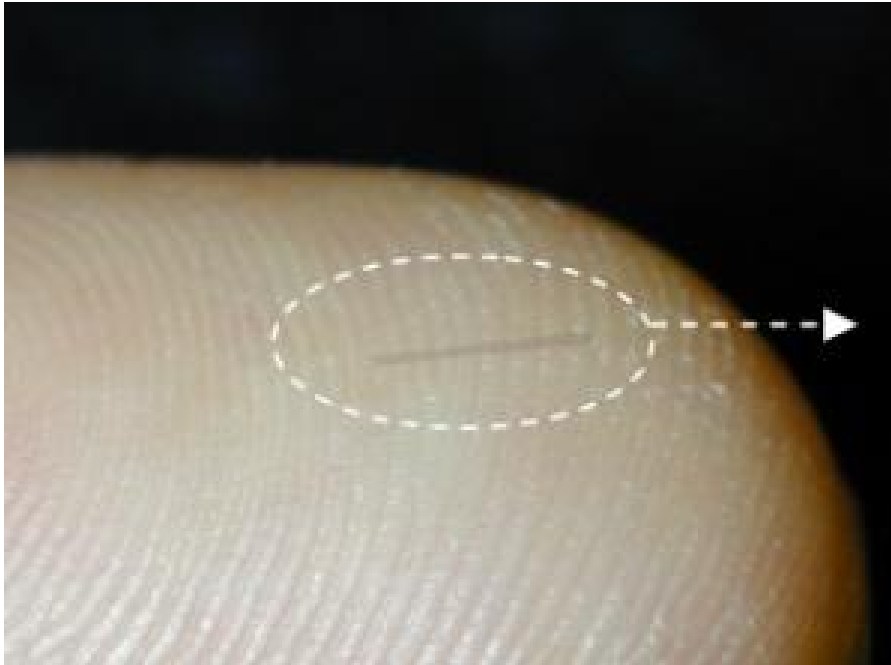
http://sci-toys.com/scitoys/scitoys/light/cliplead_transmitter.jpg

Nanotubes as Biosensors



- Carbon nanotubes can be modified to detect many molecules found in the blood
- Small size means that encapsulated nanotubes can be implanted at any site
- Researchers at University of Illinois led by Dr. Michael Strano have already constructed a glucose sensor from carbon nanotubes (<http://www.sciencedaily.com/releases/2004/12/041219153804.htm>)
- New glucose detector that takes advantage of CNT fluorescence and small size

Example: Glucose Sensor



This glass capillary tube, shown here on a fingertip, has been loaded with glucose-sensitive nanotubes. The capillary tube keeps the nanotubes confined, but has porous walls so that glucose molecules can get to them. (Credit: Michael S. Strano)

From <http://www.sciencedaily.com/releases/2004/12/041214081957.htm>

- Carbon nanotube modified with “sensor” molecules
- RESULT: More glucose → More fluorescence

Another Exciting Property of CNTs



- Nanotubes make excellent filters
 - Small pore size
 - High Surface Area to Volume Ratio
 - Easy to Modify
- Seldon, Technologies (Windsor, VT) has created several nanotube filters

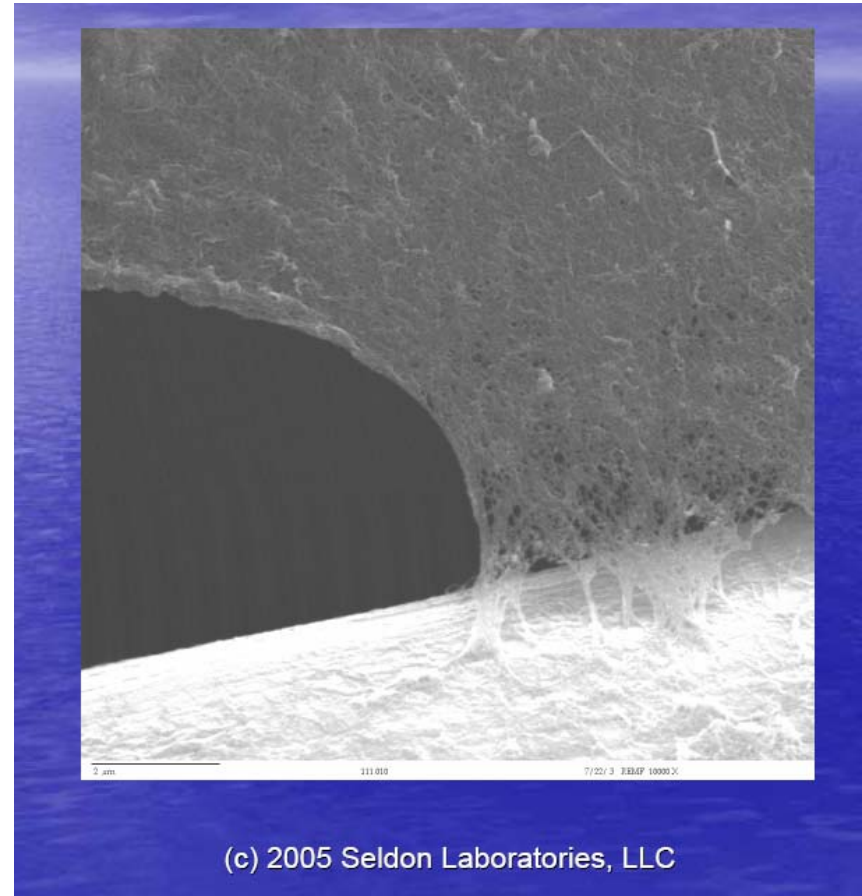
<http://www.seldontech.com/>

- Removes chemicals
- Kills and removes bio-contaminants

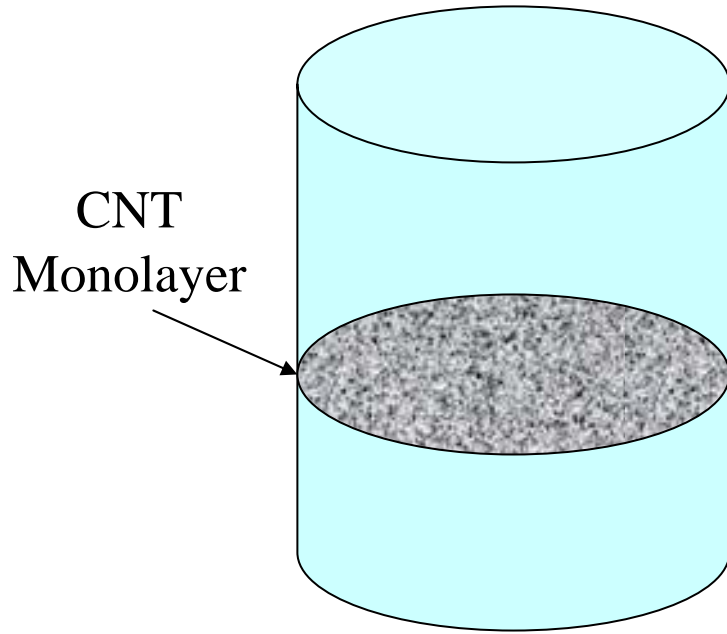
Carbon Nanotube “Kill-zone”



- Seldon Technologies created a CNT-based filter:
 - Don't need high pressure
 - “Kill-zone” that is an absolute barrier to biocontaminants
- Monolayer nanomesh blocks debris without stopping water flow



Making Your Own Monolayer

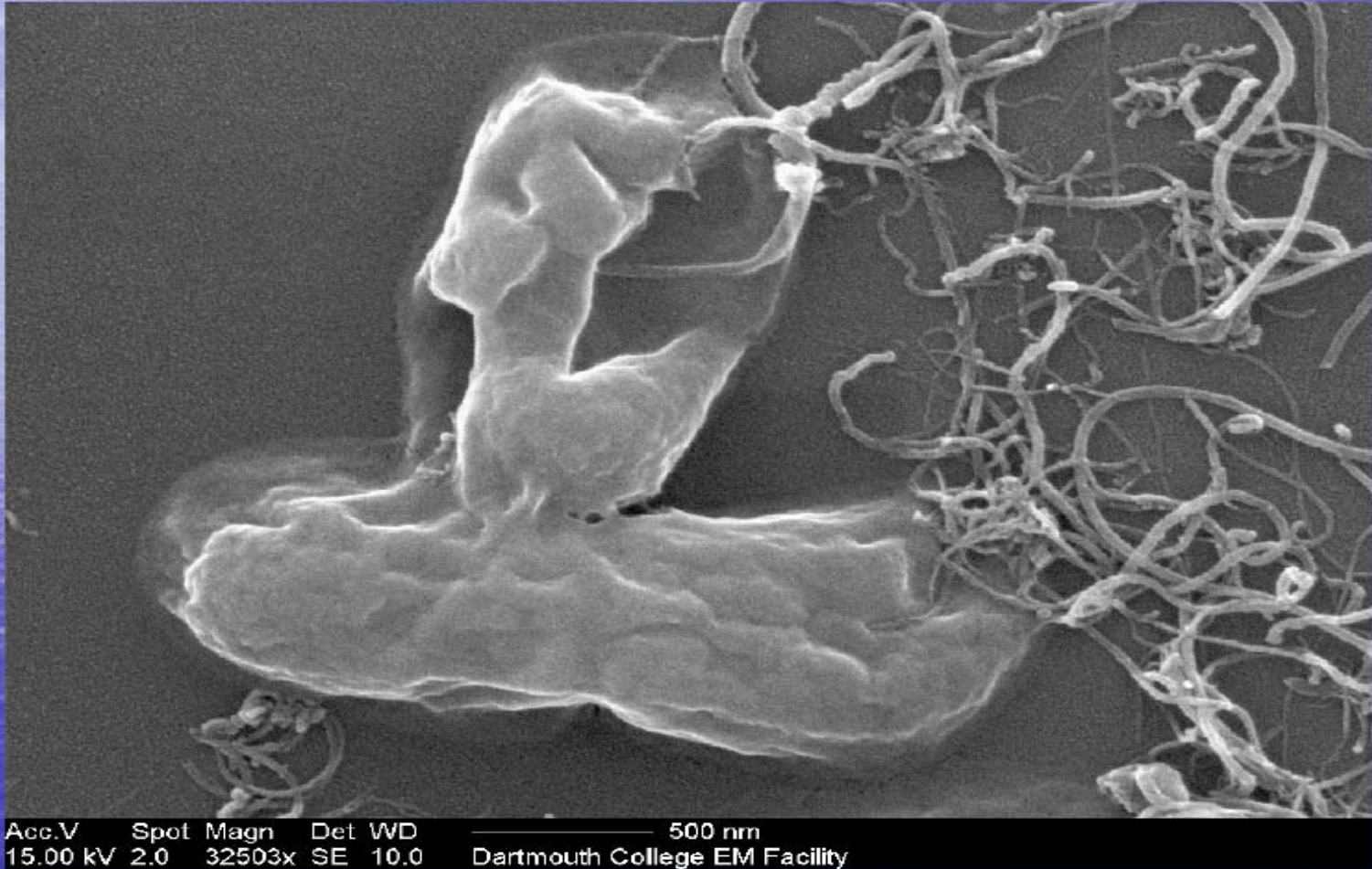


Nanotubes	Bacteria	Ratio
60 nm	2 μm	0.03
Sticks	Marbles	Ratio
1 mm	1.25 cm	0.08

TUMBLE game (\$9.99)

<http://www.pressmantoy.com/>

Antibacterial Properties



(c) 2005 Seldon Laboratories, LLC



Applications for the Military

- Prototype built as a “backpack”
- United States Air Force currently testing the device
- Can filter large volumes of water from dirty sources (EVEN URINE!)



Large Volume
Portable Field Unit

(c) 2005 Seldon Laboratories, LLC

Safe Water Available Anywhere

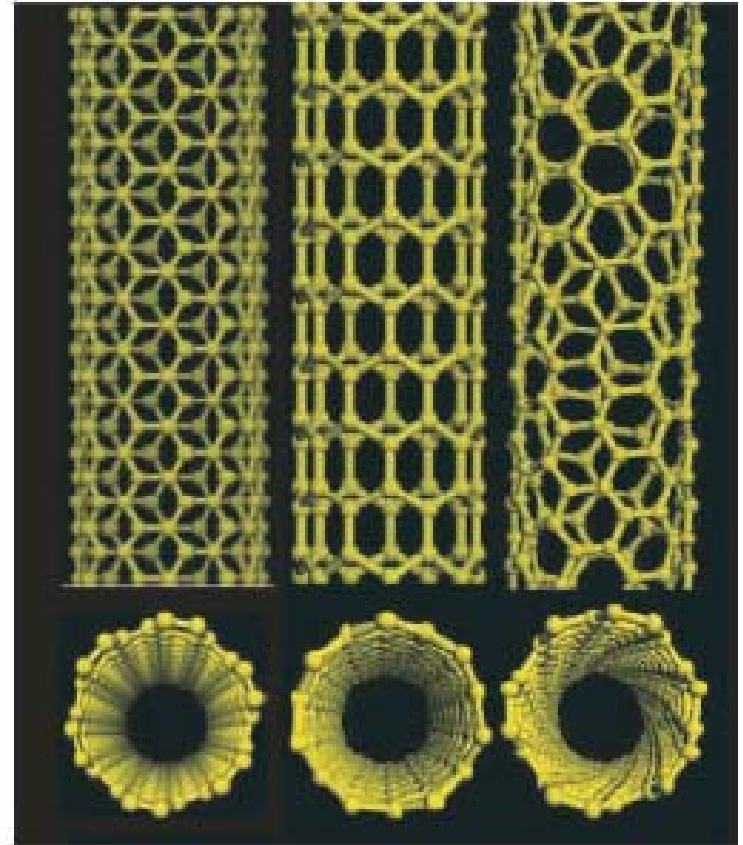


(c) 2005 Seldon Laboratories, LLC

Physics of Carbon Nanotubes



- Maximum Tensile Strength
~ 30 GPa
- Density normalized strength of a carbon nanotube is ~56 times that of steel wire
- CNT can carry large currents with little heating



Source: Baughman, R. H., Zakhidov, A. A., Heer, W. A. Carbon Nanotubes-the Route Toward Applications. *Science's Compass Review. Science*, **297**, 787-792 (2002).

What are Composite Materials?

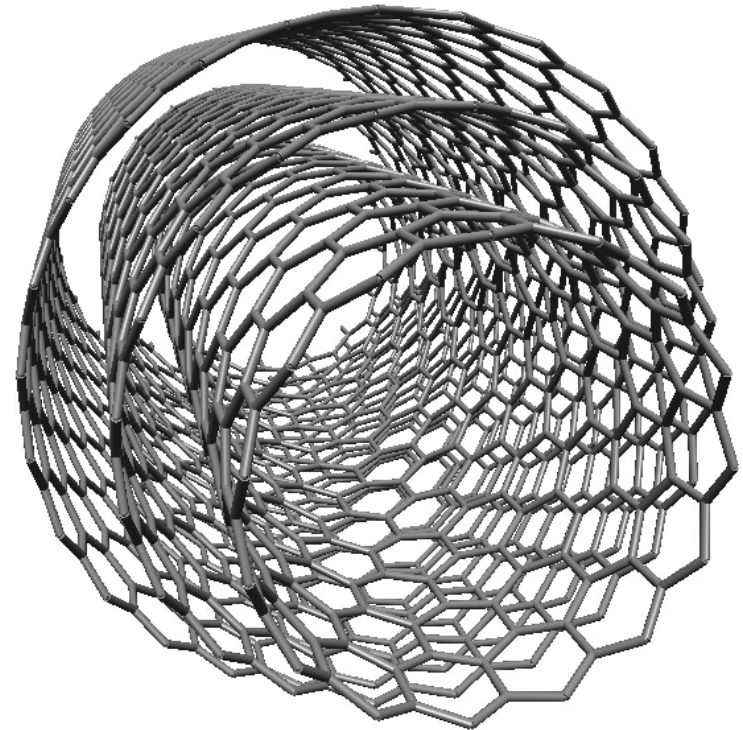


- Composite materials are materials made from two or more components
- A composite material usually consists of a strong fiber and a matrix (which surrounds the fiber)
- The fiber carries the load while the matrix keeps fibers in the correct position and protects the fibers from environmental damage.
- Examples of fibers: fiberglass, Kevlar
- Examples of a matrix: polyester, epoxy

Carbon Nanotubes in Composite Materials



- CNT can act as a strong fiber to improve the strength, elasticity, and conductivity of a material.
- Challenges
 - ❑ Dispersing the nanotubes uniformly throughout the matrix
 - ❑ Reducing intratube sliding in MWNT
 - ❑ Creating adhesion between the carbon nanotube and the matrix that results in an effective stress transfer



Mechanical Properties and applications of Carbon nanotubes (CNTs)



- Mechanical properties: (1nm to 100s nm)
 - Light weight, sustains extremely high tension force; 130 GPa compared to steel at <5 GPa
 - Highly flexible, even under low temperature.
- Applications: AFM tips, super-strong fabrics, polymer composites and space elevator



CNT in Action



Applications of CNT Composites

- Babolat Tennis Rackets
- Easton uses CNT in baseball bats
- Nissan's X-Trail SUV has bumpers that are reinforced with CNT
- Wings on NASA's Morphing Glider



© Sports Warehouse 2004



CRT Televisions

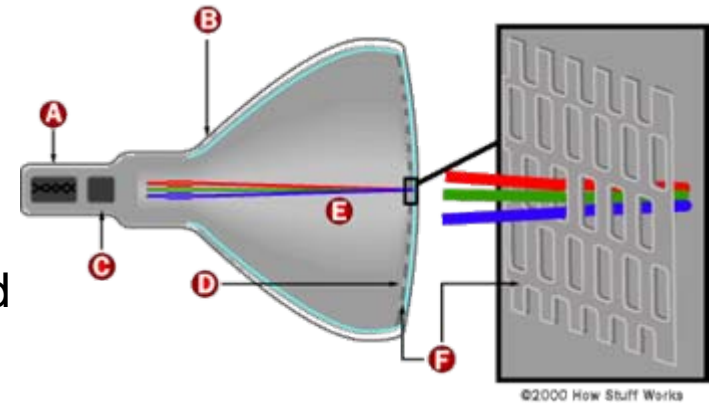


In an older TV (CRT) a cathode emits electrons towards an anode and onto a phosphor screen.

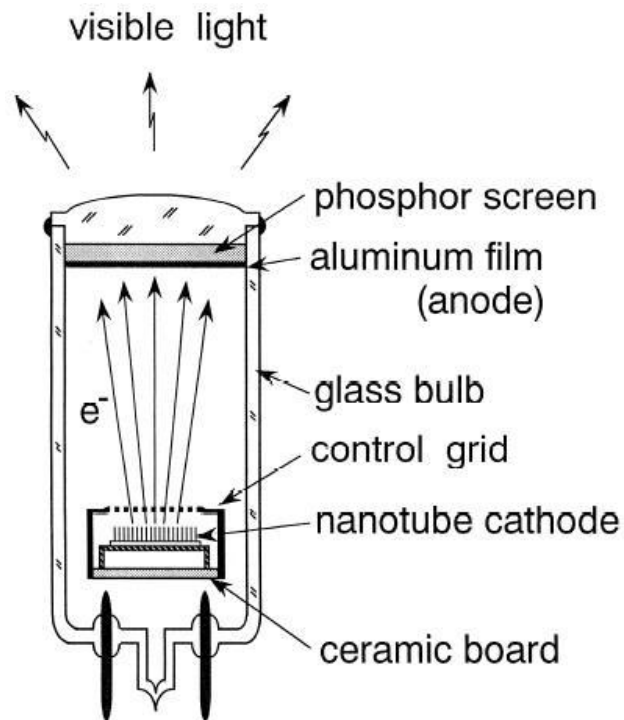
One electron beam serves the entire screen and the direction of the beam was manipulated by magnetic coils.

How does a Field Emission Display differ from a CRT?

An FED has electron emitters behind every pixel.

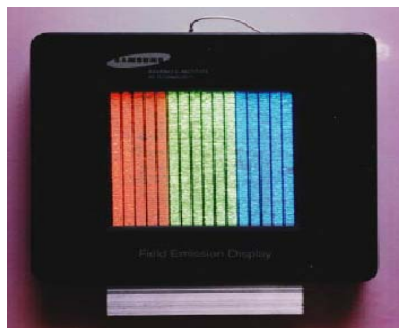


Field Emission Displays



How is an FED different from an LCD?

Lower power consumption, higher brightness, wider viewing angle.

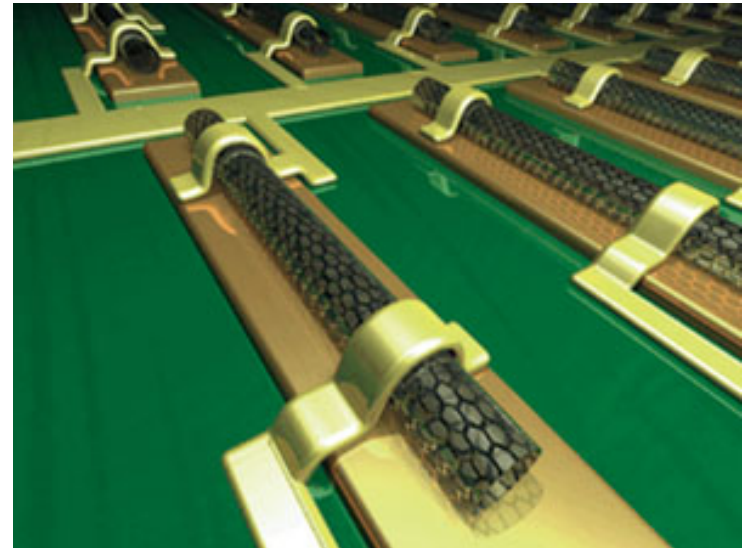
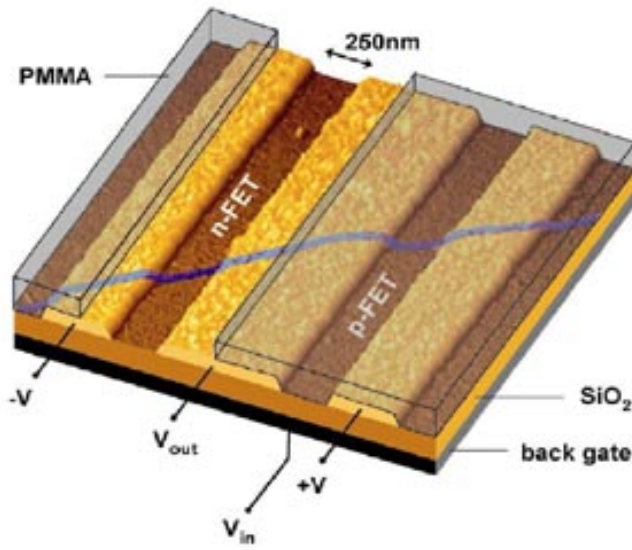


Why are carbon nanotubes a good choice in an FED?

Nanotubes are able to generate a high electric field due to their small tip. They are stable emitters with a long lifetime.

Nanoscale electronics

- Transport properties:
 - High thermal conductivities w/o electrical conductivity, Semiconducting or metallic tubes,
 - High current density (1000X Cu).
- Applications:
 - field emission devices => field emission flat panel.
 - single molecular transistors



Acknowledgments



- Boston University GK-12 program
STAMP: Science, Technology and
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- Seldon Technologies for use of their
photographs and slides
- Boston University Center for
Nanoscience and
Nanobiotechnology



CENTER FOR
NANOSCIENCE AND
NANOBIOTECHNOLOGY

The Space Elevator

