

## Lecture 8 (9/25/20)

- Reading: Ch3; 97-102
- Problems: Ch3 (text); 18, 19, 20, 23  
Ch3 (Study guide); **9**  
Ch4 (Study guide); **3**

### NEXT

- Reading: Ch4; 119-122, 125-126, 131-133  
Ch4; 123-124, 130-131, 133, 137-138
- Problems: Ch4 (text); 2, 3, 4, 8, 13, 14

## Lecture 8 (9/25/20)

### OUTLINE

#### I. Protein Structure

##### A. Primary

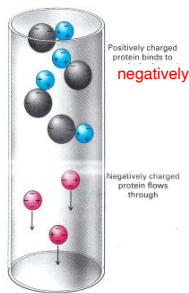
##### 1. Determination

- a. Sequence determination; CHEMICAL
  - i. Cleavage of peptides bonds
  - ii. Amino acid composition and stoichiometry
  - iii. Disrupt and determine number of chains;
  - iv. Divide & Conquer;
  - v. Edman Degradation
- b. Sequence determination; PHYSICAL
  - i. Mass Spectrometry for proteins
  - ii. Use of tandem MS/MS for sequence determination
  - iii. Isolation of proteins by 2D PAGE; Isoelectric focusing x SDS-PAGE
- c. Sequence determination; BIOLOGICAL
  - i. Genome sequenced
  - ii. Bioinformatics to predict protein sequences in predicted genes
  - iii. Use of CHEMICAL and/or PHYSICAL methods to get partial sequence

## Determination of primary structure

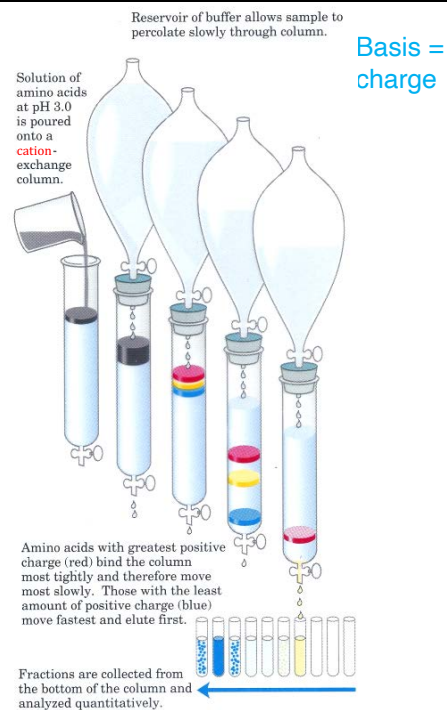
- 1) Purify protein
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- 3) Disrupt structure ( $2^\circ$ ,  $3^\circ$ ,  $4^\circ$ , and disulfides)
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- 7) Determine overlaps and piece original sequence back together

## Ion Exchange Chromatography: Example of amino-acid mixture



If this were a mixture of G, K, and D, what would the order of elution be?

1<sup>st</sup> = D, 2<sup>nd</sup> = G, 3<sup>rd</sup> = K

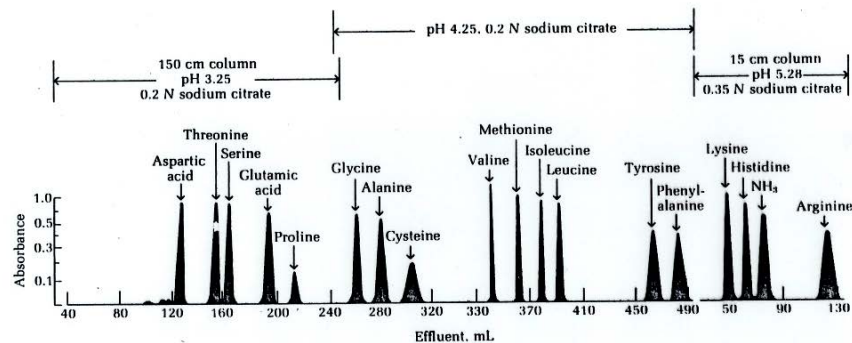


# Determination of primary structure

Wm H. Stein  
Stanford Moore

## Amino Acid Composition

Figure 5-15  
Automatically recorded chromatographic analysis of amino acids on a cation-exchange resin. The elution is carried out with different buffers of successively higher pH. The effluent is caught in small volumes, and the amino acid content of each tube is automatically analyzed. The area under each peak is proportional to the amount of each amino acid in the mixture.



Absorbance is at 570 nm (purple). How did they make the amino acid residues colored? .....ninhydrin

How many AA are "resolved?" .....18

Where are the other 2? .....Asn & Gln have already been converted to Asp & Glu, respectively

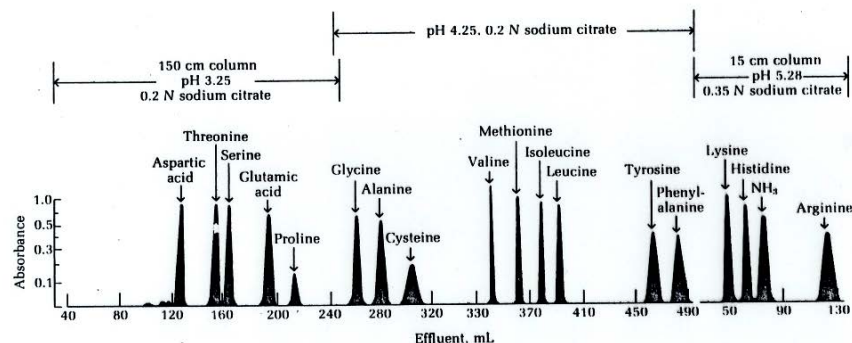
# Determination of primary structure

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## Amino Acid Composition

Figure 5-15  
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Cation exchanger=Dowex-50 (sulfonic acid on polystyrene)



Calculate the "mole%" or "mole fraction"; in other words the stoichiometry:

Example; protein MW = 11,000 Da

Approximate #AA =  $11,000/110$  (Ave MW for AA) = 100 AA

If 10% of  $A_{570}$  is 10% for Arg:  $100 \times 0.10 = 10$  Arg in protein

## Determination of primary structure

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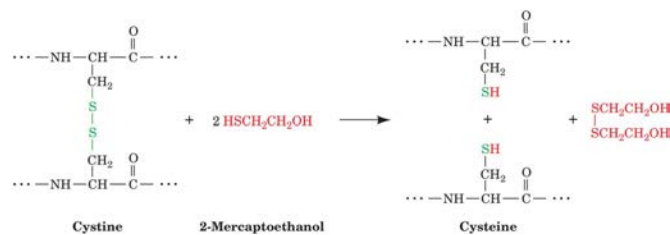
38

## Disrupt structure (2°, 3°, 4°, and disulfides)

What holds these levels of structure together? .....non-covalent bonds (H-bonds, van der Waals, ionic, hydrophobic)

What have you used in the lab that might disrupt non-covalent bonds? .....Urea, SDS, pH extremes, heat, etc.

What about the covalent S-S bond? .....2-mercaptoethanol (β-mercaptoethanol, BME) or dithiothreitol (DTT).



To keep disulfides from reforming.....

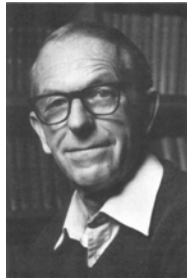
- 1) keep BME at high concentration in buffers
- 2) alkylate the SH groups

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## Determine the number of peptide chains by counting number of amino terminal ends



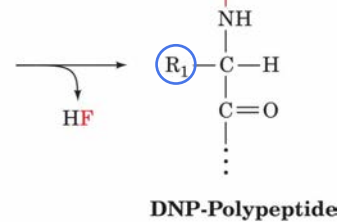
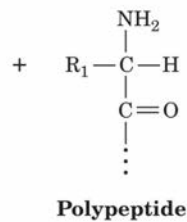
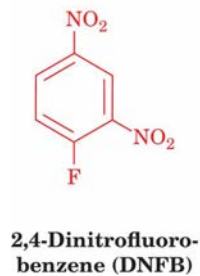
Frederick Sanger  
(1918-2013)

### Sanger's Reagent

Example:

small tripeptide

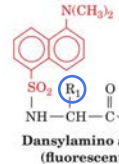
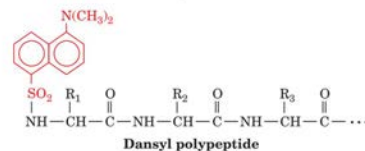
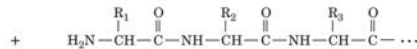
- AA comp = A,K,L
- C-term = K
- DNP-A
- Sequence is A-L-K



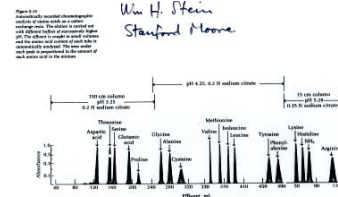
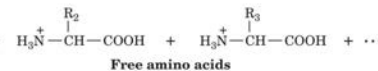
Absorbs at 353 nm (yellow)

Fig 3-26

## Determine the number of peptide chains by counting number of amino terminal ends



Excitation at 330 nm  
Emission at 519 nm



Wm H. Stein  
Stanford Moore

## Determination of primary structure

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## Divide into fragments:

## Proteolytic Cleavage

TABLE 5-4 Specificities of Various Endopeptidases

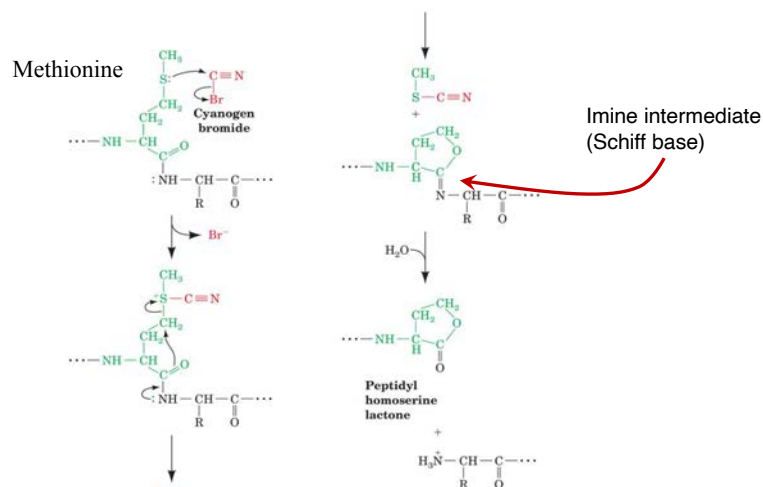
$\begin{array}{c} R_{n-1} \quad O \quad R_n \quad O \\   \quad    \quad   \quad    \\ -NH-CH-C-NH-CH-C- \\ \uparrow \\ \text{Scissile peptide bond} \end{array}$			
Enzyme	Source	Specificity	Comments
Trypsin	Bovine pancreas	$R_{n-1}$ = positively charged residues: Arg, Lys; $R_n \neq$ Pro	Highly specific
Chymotrypsin	Bovine pancreas	$R_n$ = bulky hydrophobic residues: Phe, Trp, Tyr; $R_{n-1} \neq$ Pro	Cleaves more slowly for $R_{n-1}$ = Asn, His, Met, Leu
Elastase	Bovine pancreas	$R_{n-1}$ = small neutral residues: Ala, Gly, Ser, Val; $R_n \neq$ Pro	
Thermolysin	<i>Bacillus thermoproteolyticus</i>	$R_n$ = Ile, Met, Phe, Trp, Tyr, Val; $R_{n-1} \neq$ Pro	Occasionally cleaves at $R_n$ = Ala, Asp, His, Thr; heat stable
Pepsin	Bovine gastric mucosa	$R_n$ = Leu, Phe, Trp, Tyr; $R_{n-1} \neq$ Pro	Also others; quite nonspecific; pH optimum = 2
Endopeptidase V8	<i>Staphylococcus aureus</i>	$R_{n-1}$ = Glu	

TABLE 5-5 Specificities of Exopeptidase

$\begin{array}{c} R_{n-1} \quad O \quad R_n \quad O \\   \quad    \quad   \quad    \\ -NH-CH-C-NH-CH-C-O^- \\ \uparrow \\ \text{Scissile peptide bond} \end{array}$			
Carboxypeptidase A	Bovine pancreas	$R_n$ = C-terminal; $R_{n-1} \neq$ Pro	

## Divide into fragments:

## Cyanogen Bromide Cleavage



Similar result as an endopeptidase, but leaves a **homoserine lactone** as the C-terminal residue

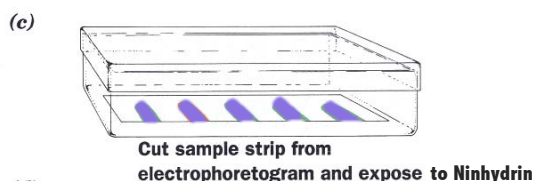
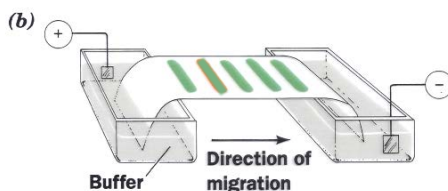


## Separation and isolation of peptide fragments



Example: paper electrophoresis

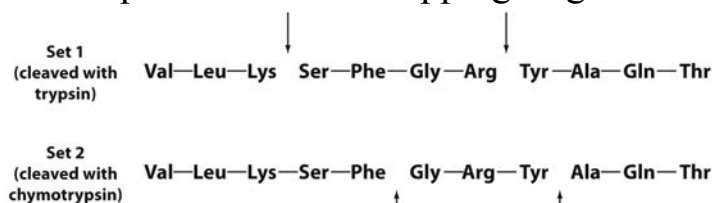
Also, TLC, silica gel, etc.



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## Determination of primary structure

- Determine amino-acid composition
- Dansyl chloride or FDNB to determine amino-termini and number
- Proteases: Cleaves peptide bonds only after specific residues.
- **Cleave protein with 2 different proteases.**
- Sequence fragments with **Edman degradation**. Piece together sequence from overlapping fragments.



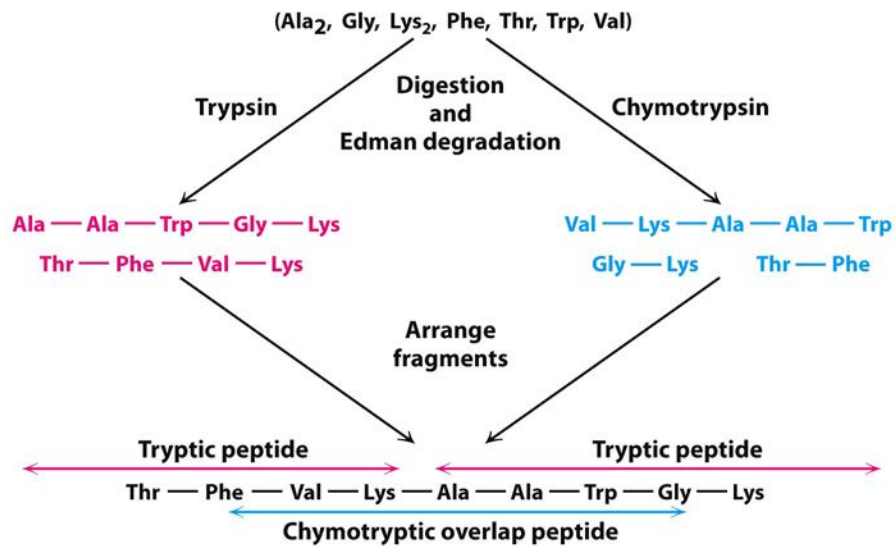
48

## Determination of primary structure: Divide & Conquer

(Ala<sub>2</sub>, Gly, Lys<sub>2</sub>, Phe, Thr, Trp, Val)

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## Determination of primary structure: Divide & Conquer



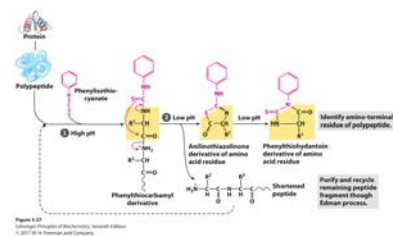
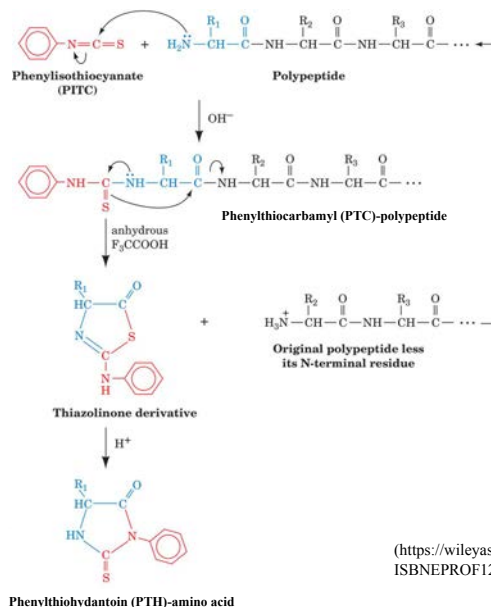
50

## Determination of primary structure

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## Determine the Sequence: Edman degradation



Edman Degradation Animated Figure

([https://wileyassets.s3.amazonaws.com/Voet\\_Fundamentals\\_of\\_Biochemistry\\_Se\\_ISBNPROF12533/media/animated\\_figures/ch05/5-15.html](https://wileyassets.s3.amazonaws.com/Voet_Fundamentals_of_Biochemistry_Se_ISBNPROF12533/media/animated_figures/ch05/5-15.html))

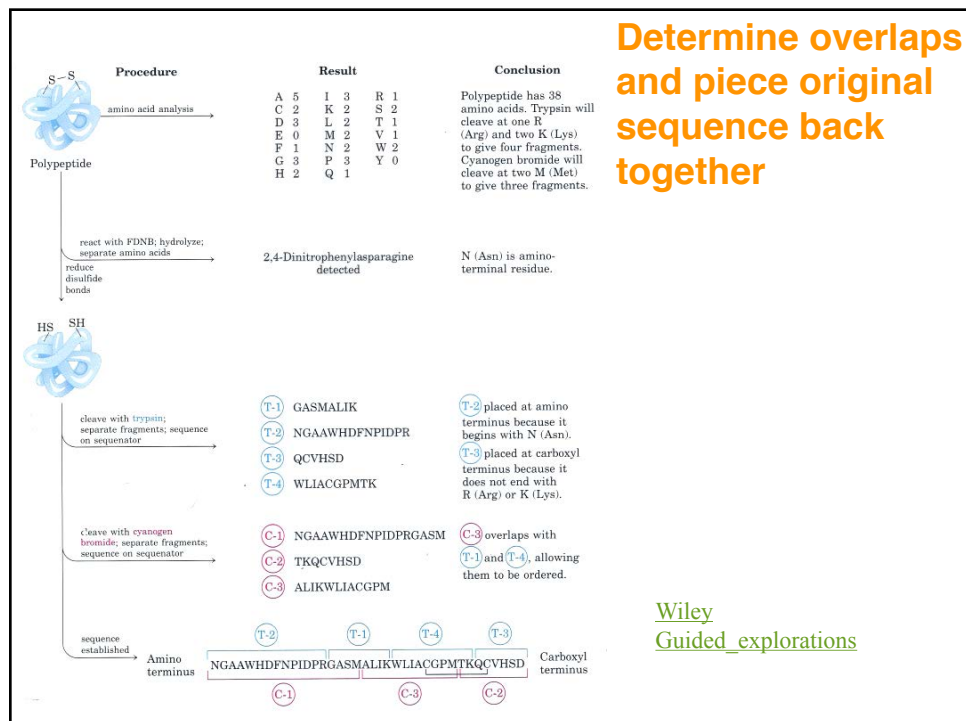
Copyright © 2016 John Wiley & Sons, Inc. All rights reserved. Figure 5-16

WILEY

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## Determination of primary structure

THREE basic ways to know the primary structure. Only the CHEMICAL method will give the entire covalent structure, including any disulfide bonds. But other methods are more sensitive. One can classify these methods by:

**CHEMICAL**  
**PHYSICAL**  
**BIOINFOMATICAL**

We just went through the CHEMICAL.

The PHYSICAL method still requires the same strategy, including purification, fragmentation, chromatography, and alignment.

But, instead of an Edman degradation the use of tandem Mass Spectrometry (MS) is employed.

Lets look at the use of MS in biochemistry

- Ions “fly” in a vacuum toward a target with a velocity  $\propto z/m$  (charge-to-mass ratio)
- Molecules with higher charge and lower mass get detected first.
- Molecules with a lower charge and higher mass get detected last.
- Plotted as  $m/z$  to read peaks from left to right
- Instruments can distinguish molecules with same charge by  $< 1$  Da

## Determine the Sequence: Tandem MS

The major problem in using MS for macromolecules is getting them to “fly” in a vacuum with a charge.

TWO major methods:

- 1) Electro-Spray Ionization (ESI)
- 2) Matrix-Assisted Laser-Desorption Ionization (MALDI)

**ESI**

$$\text{Mass} = (m/z) \times z$$

$$(m/z_A) \times z_A = (m/z_B) \times z_B$$

$$893.3z = 848.7(z+1)$$

$$893.3z = 848.7z + 848.7$$

$$848.7 = z(893.3 - 848.7) = z44.6$$

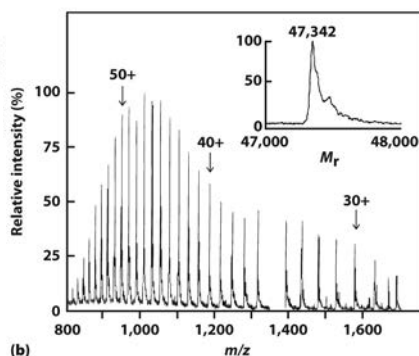
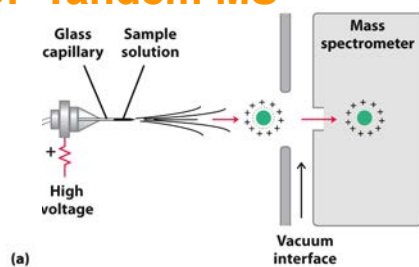
$$848.7/44.6 = 19 = z$$

$$\text{Mass} = 893.3 \times 19 = 16,973$$

$$\text{Mass} = 848.7 \times 20 = 16,974$$

$$\text{Mass} = 1696.3 \times 10 = 16,963$$

**What is MALDI?**



Information from M. Mann and M. Wilm, Trends Biochem. Sci. 20:219, 1995

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$$893.3z = 848.7(z+1)$$

$$893.3z = 848.7z + 848.7$$

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**What is MALDI?**

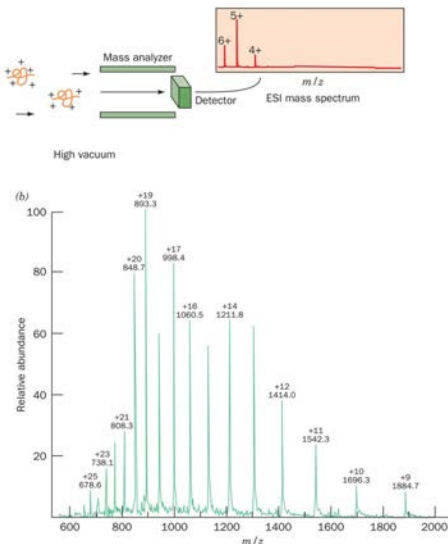
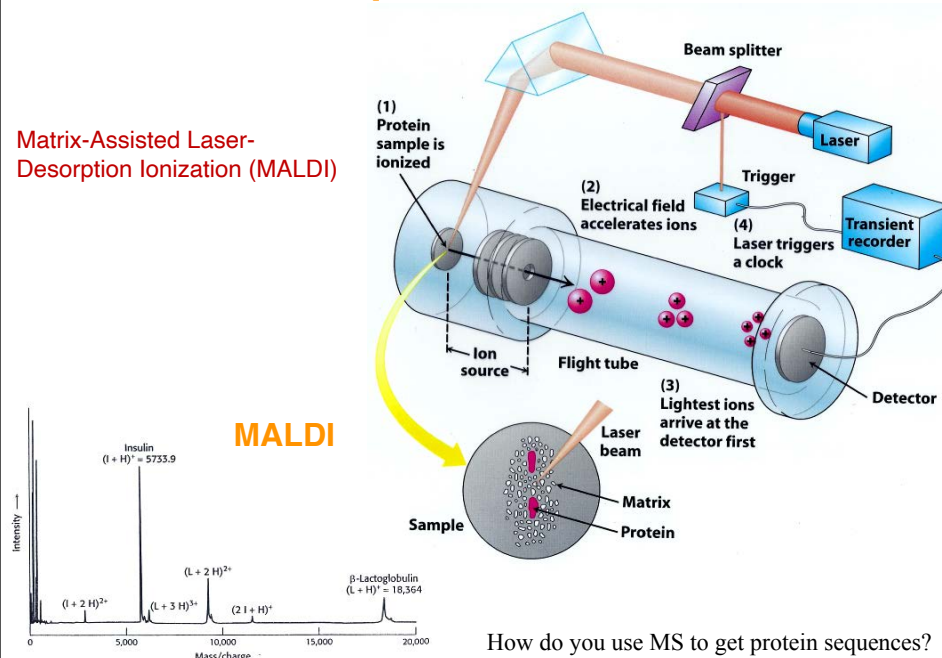


Figure 5-17

## Determine the Sequence: Tandem MS

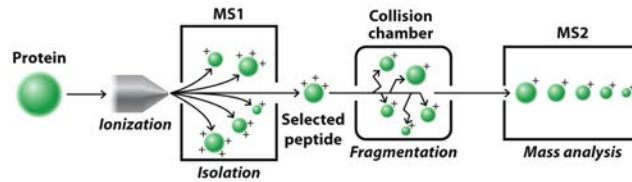
**Matrix-Assisted Laser-Desorption Ionization (MALDI)**

**MALDI**



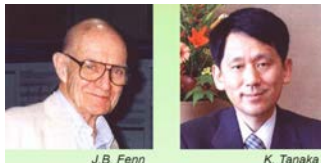
How do you use MS to get protein sequences?

## Determine the Sequence: Tandem MS

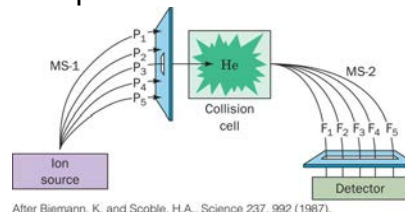


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- Mass spectrometry uses mass-to-charge ratio of different ions to determine mass
- Tandem MS-MS: First selects a peptide, then fragmentation, and second determines mass of fragments
- By comparing all of fragments, those that different by mass of one amino acid to determine sequence



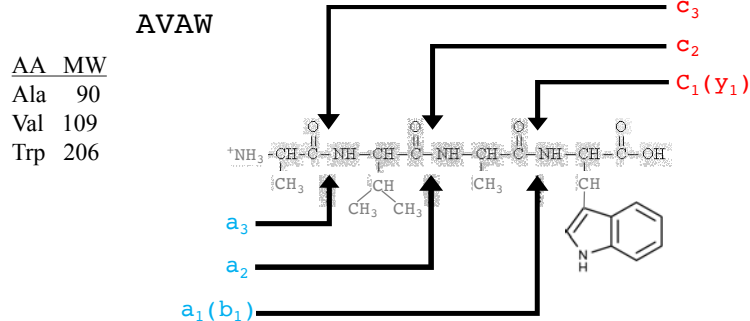
Nobel Prize in Chemistry 2002



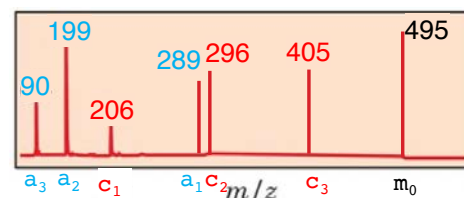
After Riemann, K. and Scoble, H.A. Science 237, 992 (1987)

## Determine the Sequence: Tandem MS

- EXAMPLE



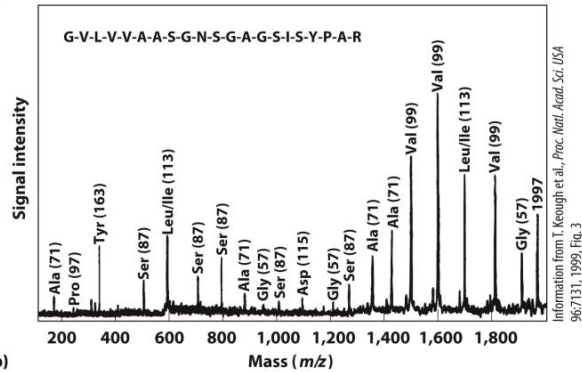
AVAW



## Determine the Sequence: Tandem MS

### • EXAM

AA	MW
Ala	90
Val	109
Trp	206

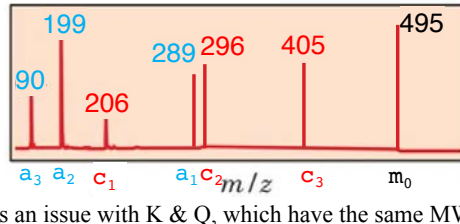
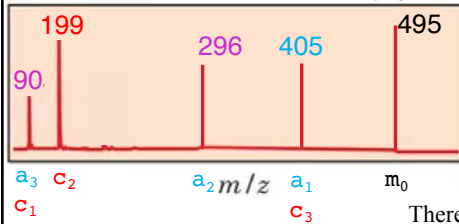


(b)

Figure 3-31

Lehninger Principles of Biochemistry, Seventh Edition

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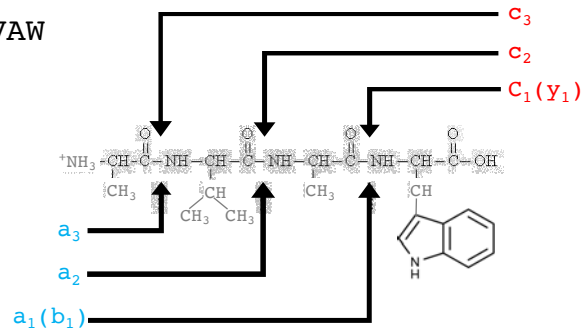
There is an issue with K &amp; Q, which have the same MW!

## Determine the Sequence: Tandem MS

### • EXAMPLE

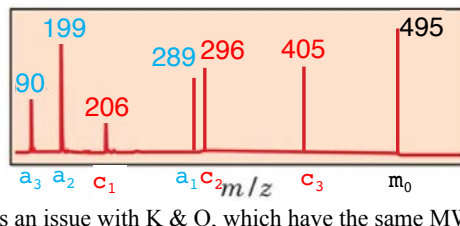
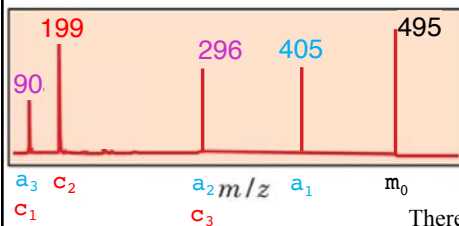
AA	MW
Ala	90
Val	109
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AAVAV



AAVVA

AAVAV



There is an issue with K &amp; Q, which have the same MW!