Lecture 31 (12/07/20)		Carbohydrates			
TODAY		A. L B. F	Jet Rol	es	
•Reading:	Chs4,6,8,10,14,16,17,18; 128-129,	C. N	No	nosaccharides	
•Problems:	189,311,377-380,555-557,561,621-622,639,662-663,679 691-694	,	1.	Chirality a. One or more asymmetric carbons	
Problems.	-		2	D. Linear and ring forms	
NEXT (LAST	-ii)		۷.	carbohydrates	
•Reading:	Ch13; 497-501, 507-514		3.	Polymerization a. The Glycosidic Bond	
•Problems:	Ch13 (text); 5,6,9,10,22,24	D. (	Olig	gosaccharides	
	Ch7 (study-guide: applying); 4		1.	Glycoproteins & glycolipids	
	Ch7 (study-guide: facts); 1-4,10-12		2.	O-linked	
			3. 4	Sequence determination-ABO	
		E. F	Pol	vsaccharides	
•Additional Office hours – Thursday (12/10) & Wednesday (12/16) at 10:00-11 am		:	1. 2.	Polymers of glucose Polymers of disaccharides	
•Review session – next Monday 2-3 PM SCI109		Vita	am A.	Nins & Cofactors Water soluble	
•Final Exam – Th	nursday <mark>8</mark> -11 AM in LAW	Me	tał	polism	



# **Polysaccharides**

## Carbohydrates

#### **Polysaccharides**

- The majority of natural carbohydrates are usually found as large polymers.
- These polysaccharides can be:
  - homopolysaccharides (one monomer unit)
  - heteropolysaccharides (multiple monomer units)
  - linear (one type of glycosidic bond)
  - branched (multiple types of glycosidic bonds)
- Polysaccharides do not have a defined molecular weight.
  - This is in contrast to proteins because, unlike proteins, no template is used to make polysaccharides.
  - Polysaccharides are often in a state of flux; monomer units are added and removed as needed by the organism.

















## Polysaccharides: Polymers of Disaccharides First, need to describe the Extracellular Matrix (ECM)

- · Material outside the cell
- · Strength, elasticity, and physical barrier in tissues (varies tremendously)
- Main components

   proteoglycans
   collagen & elastin fibers

   Proteoglycans

   Different glycosaminoglycans are O-linked to the "core protein."
   Linkage from anomeric carbon of xylose to serine hydroxyl
   Our tissues have many different core proteins; aggrecan is the best studied.



#### Polysaccharides: Polymers of Disaccharides Glycosaminoglycans

(the carbohydrate part of proteoglycans)

- Linear polymers of repeating disaccharide units (sugarX-sugarY)<sub>n</sub>
  - One monomer (sugarX) is either sugar acid or Gal
    - uronic acids (C6 oxidation)
    - Most have sulfate esters
- One monomer (sugarY) is either:
  - N-acetyl-glucosamine (GlcNAc) or N-acetyl-galactosamine (GalNAc)
  - Also sulfate esters
- Extended hydrated molecule
  - Negatively charged
  - minimizes charge repulsion
- Forms meshwork with fibrous proteins to form extracellular matrix
  - connective tissuelubrication of joints
- Form huge (M<sub>r</sub> > 2 10<sup>8</sup>) noncovalent aggregates (Hyaluronan and Aggrecan).
  - They hold a lot of water (1000× its weight) and provide lubrication.
  - Very low friction material
  - Covers joint surfaces: articular cartilage
  - reduced friction & load balancing



#### Polysaccharides: Polymers of Disaccharides Glycosaminoglycans

	Usual molecular weight of poly- saccharide chain	Component sugars	Location of sulfate	Linkage	Major Source
Hyaluronic acid <sup>a</sup>	1 - 3 x 10 <sup>6</sup>	N-acetylglucosamine glucuronic acid	•	β-(1+4) β-(1+3)	synovial fluid. vitreous humor of the eye, umbilical cord, cock's comb
Chondroitin 4-sulfate (chondroitin sulfate A)	2 - 5 x 104	N-acetylgalactosamine glucuronic acid	4	β-(1+4) β-(1+3)	human cartilage, aorta
Chondroitin 6-sulfate (chondroitin sulfate C)	2 - 5 x 10 <sup>4</sup>	N-acetylgalactosamine glucuronic acid	6	8-(1+4) 8-(1+3)	heart valves
Dermatan sulfate (chondroitin sulfate B)	2 - 5 x 10 <sup>4</sup>	N-acetylgalactosamine iduronic acid glucuronic acid	4	$\beta - (1+4)$ $\alpha - (1+3)^{b}$ $\beta - (1+3)$	skin, blood vessels, heart valves
Heparin	1 - 3 x 10 <sup>4</sup>	glucosamine glucuronic acid iduronic acid	3,6,N 2	a-(1+4) B-(1+4), a-(1+4) <sup>b</sup>	lung, mast cells
Heparan sulfate (heparitin sulfate)	$2 - 10 \times 10^3$	glucosamine N-acetylglucosamine glucuronic acid iduronic acid	N ? 3,6 2	α-(1+4) β-(1+4)	blood vessels, cell surfaces
Meratan sulfate	$5 - 20 \times 10^3$	N-acetylglucosamine galactose	6	B-(1+3) B-(1+4)	cornea of the eye, nucleus pulposus, cartilage

<sup>a</sup>The attachment of hyaluronic acid to protein has not been demonstrated unequivocally.

bThis linkage of L-iduronic acid, identical to the 8-linkage of D-glucuronic acid. However, iduronic acid is of the L rather than D configuration, which results in this bond being designated as a rather than 8.



















Primer	Type <sup>a</sup>	Repeating unit <sup>b</sup>	Size (number of monosaccharide units)	<b>Roles/significance</b>
Starch				Energy storage: in plants
Amylose Amylopectin	Homo- Homo-	$(\alpha 1 \rightarrow S4)$ Glc, linear $(\alpha 1 \rightarrow S4)$ Glc, with $(\alpha 1 \rightarrow S6)$ Glc branches every 24–30 residues	50–5,000 Up to 10 <sup>6</sup>	
Glycogen	Homo-	$(\alpha 1 \rightarrow S4)$ Glc, with $(\alpha 1 \rightarrow S6)$ Glc branches every 8–12 residues	Up to 50,000	Energy storage: in bacteria and animal c
Cellulose	Homo-	(β1 <b>→</b> S4) Glc	Up to 15,000	Structural: in plants, gives rigidity and strength to cell walls
Chitin	Homo-	(β1 <b>→</b> S4) GlcNAc	Very large	Structural: in insects, spiders, crustacean gives rigidity and strength to exoskeletor
Peptidoglycan	Hetero-; peptides attached	4)Mur2Ac(β1 <b>→</b> S4) GlcNAc(β1	Very large	Structural: in bacteria, gives rigidity and strength to cell envelope
Hyaiuronan (a glycosaminoglycan)	Hetero-; acidic	4)GICA (β1 <b>→</b> 83) GICNAC(β1	Up to 100,000	structural: in vertebrates, extracellular n of skin and connective tissue; viscosity a lubrication in joints